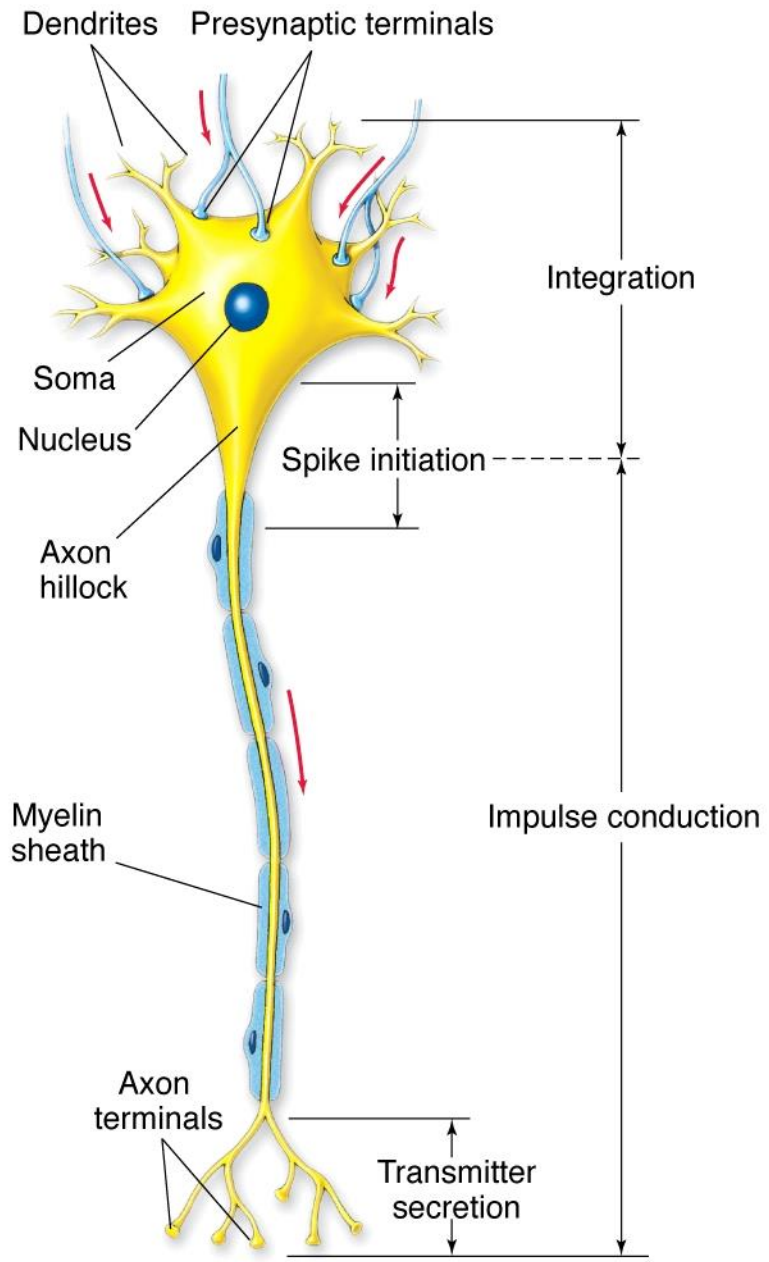
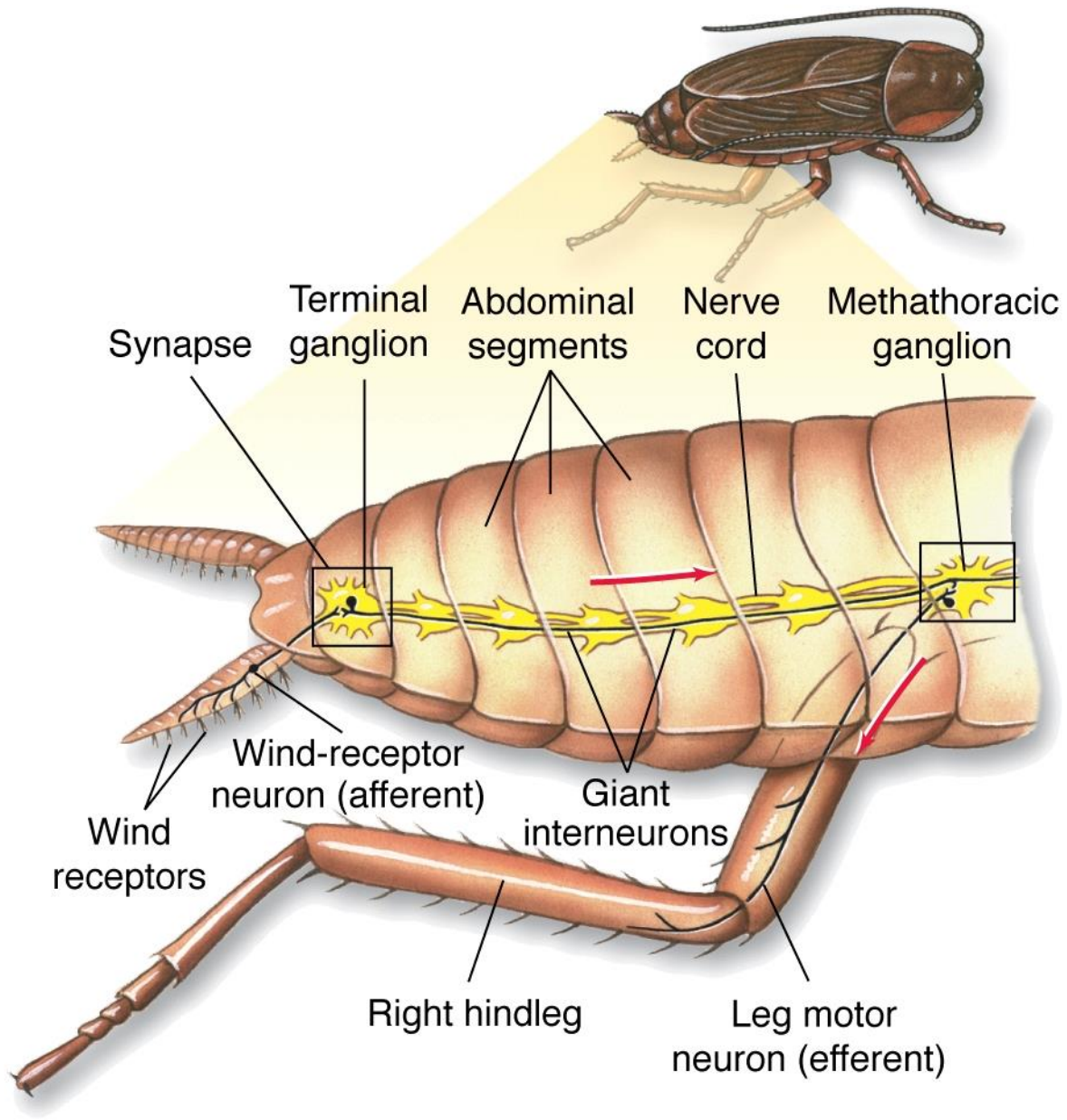
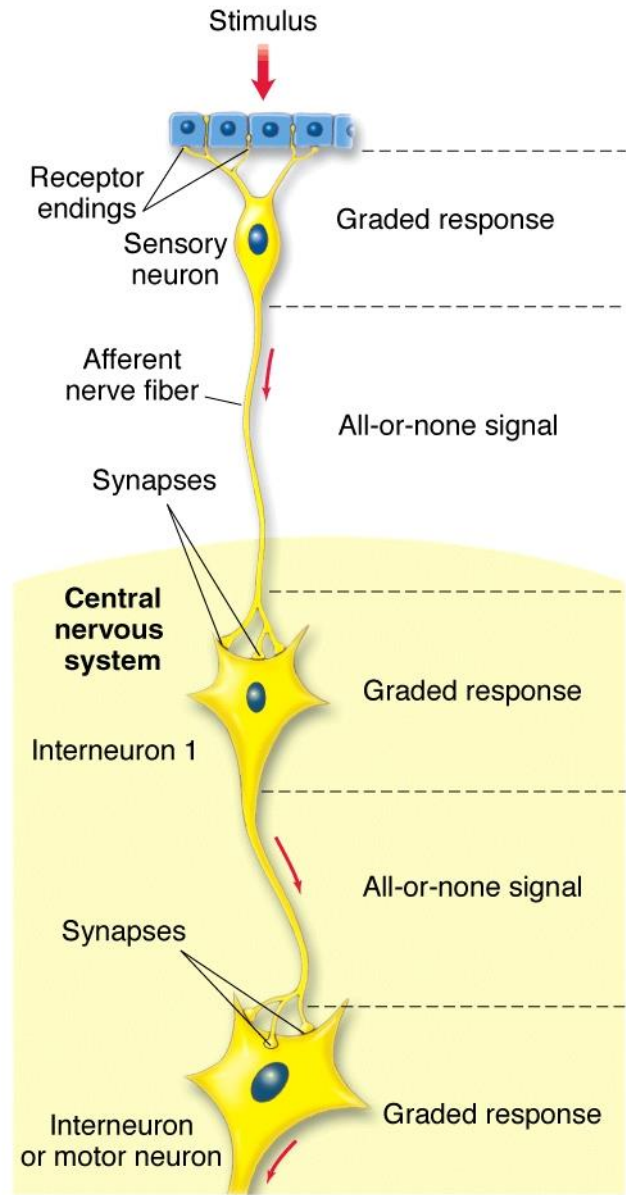


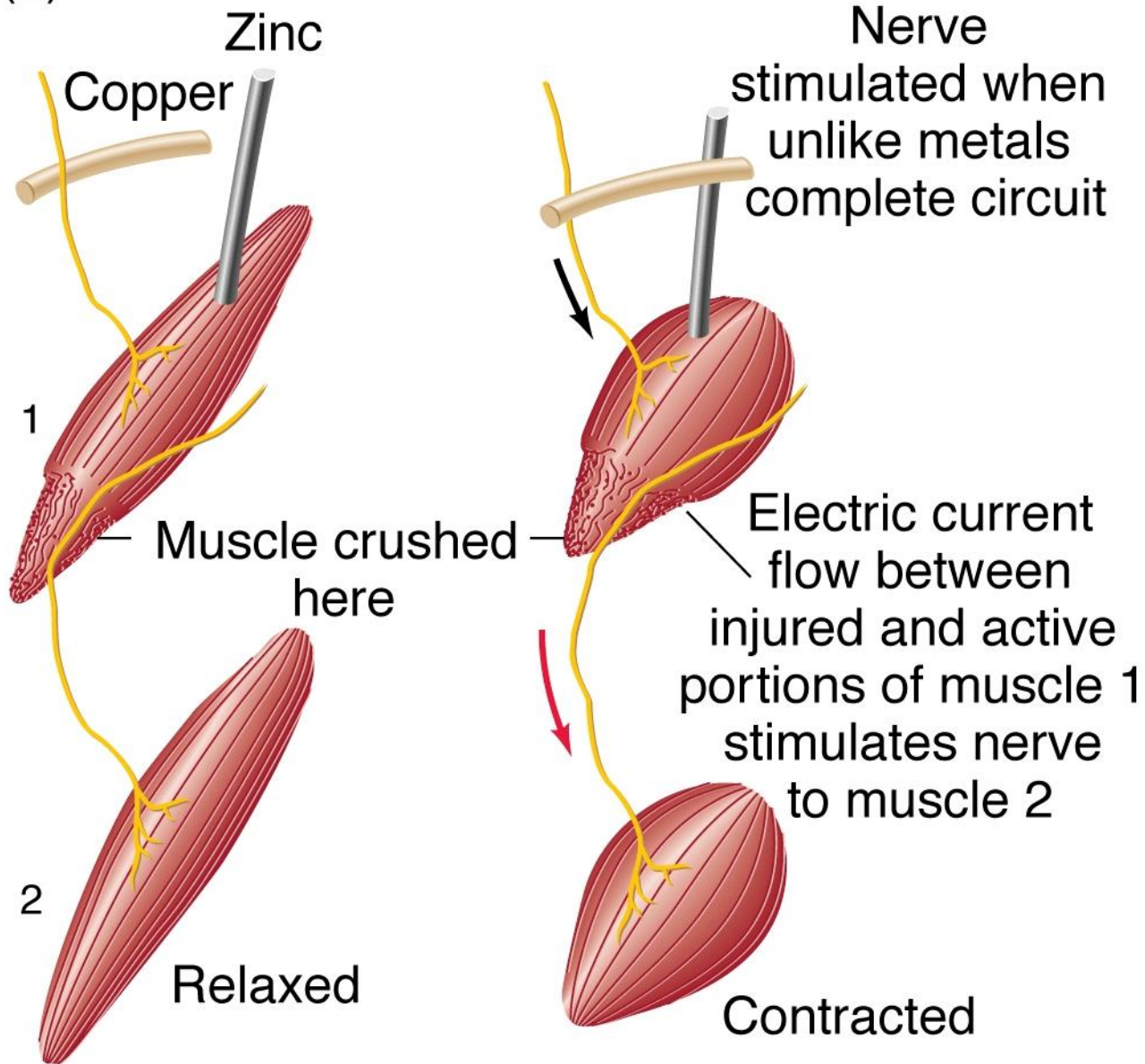
Neuronal function



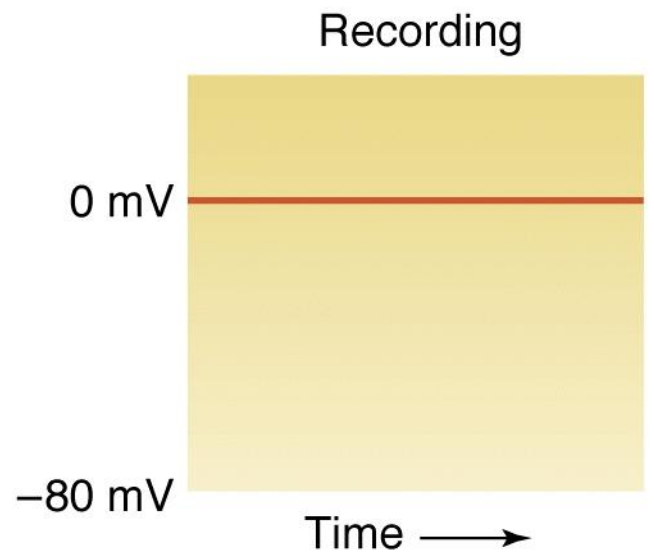
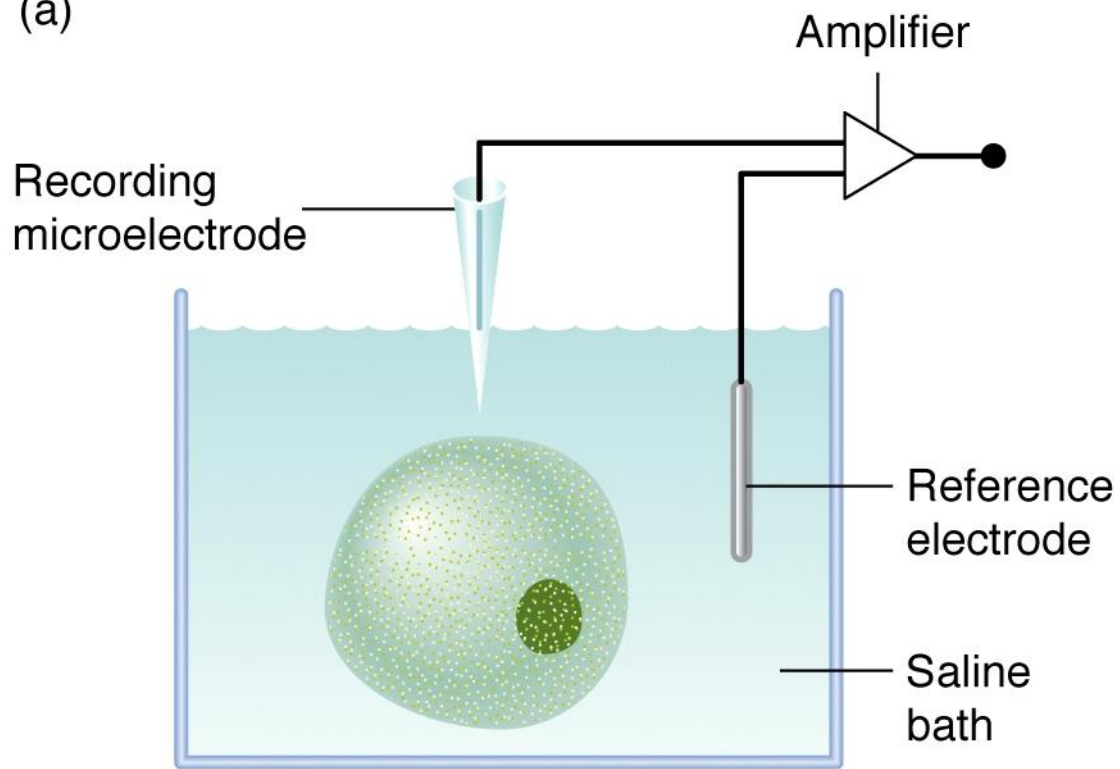




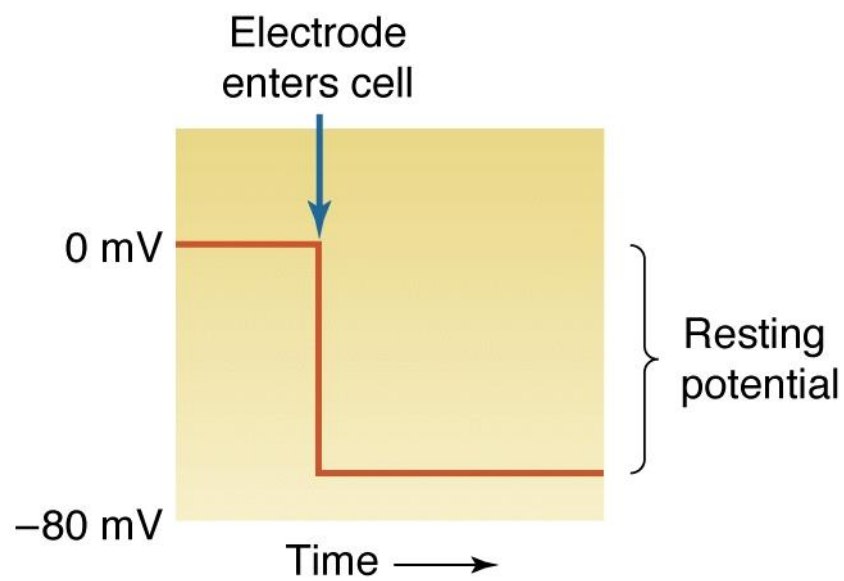
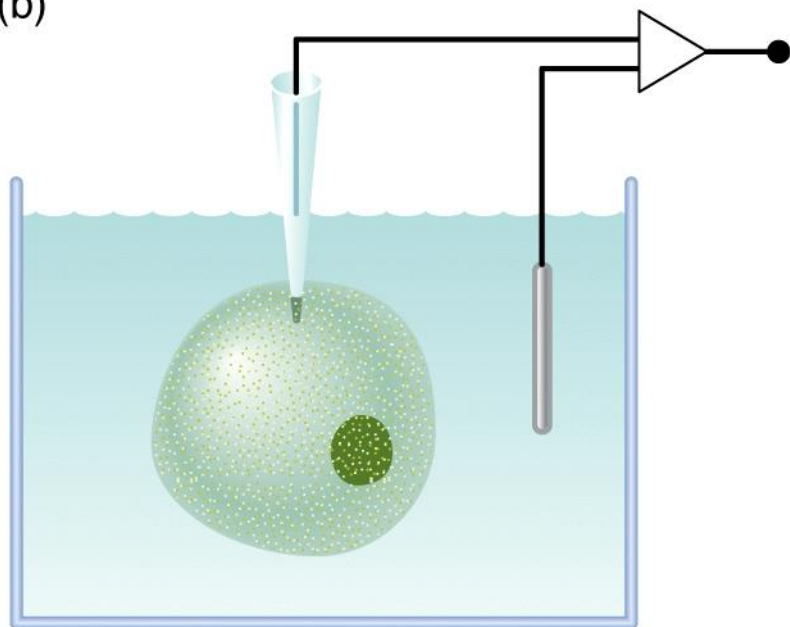
(b)



(a)



(b)



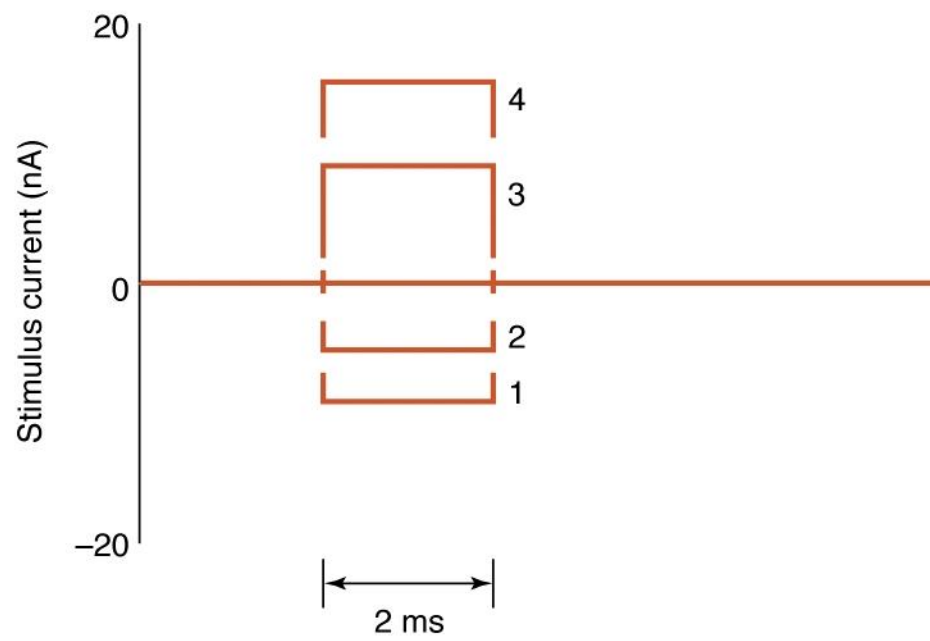
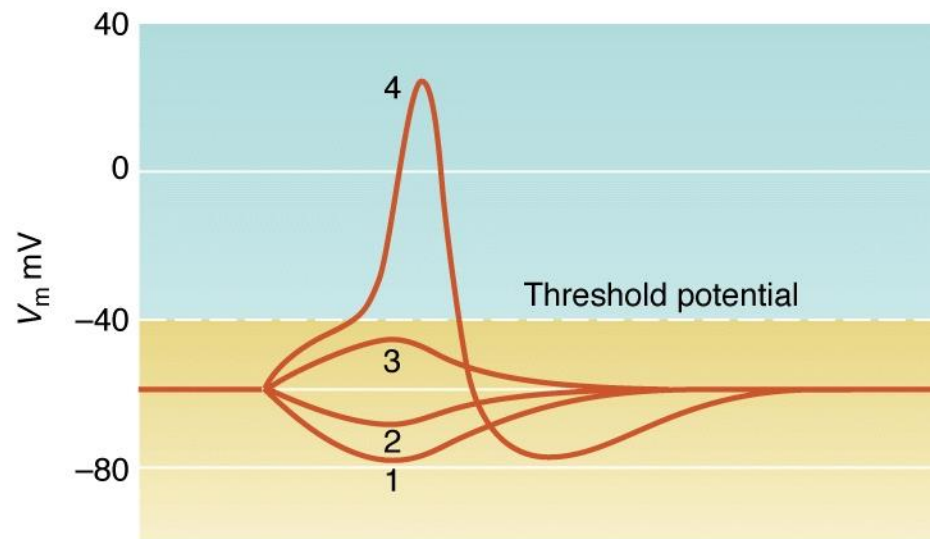
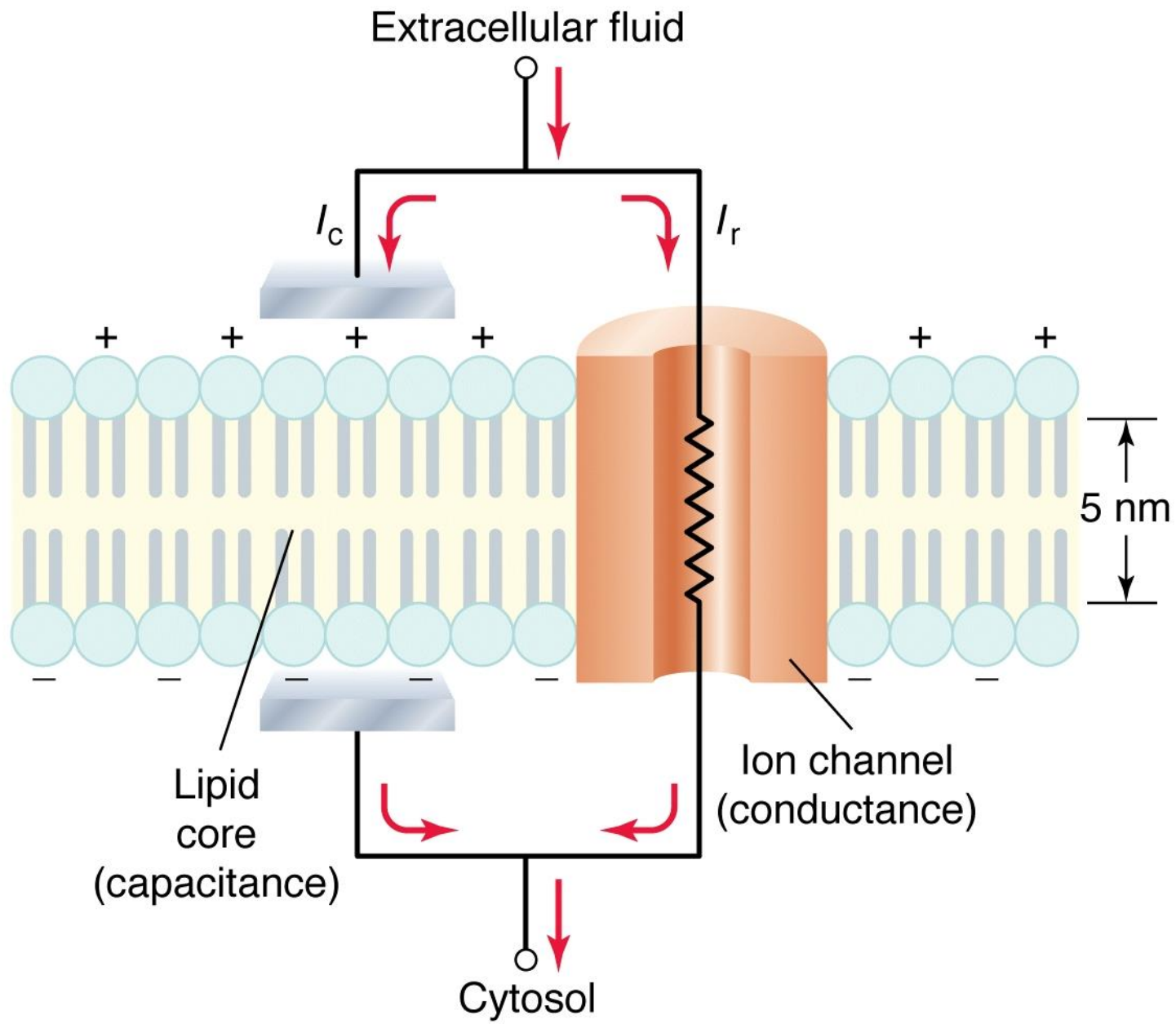
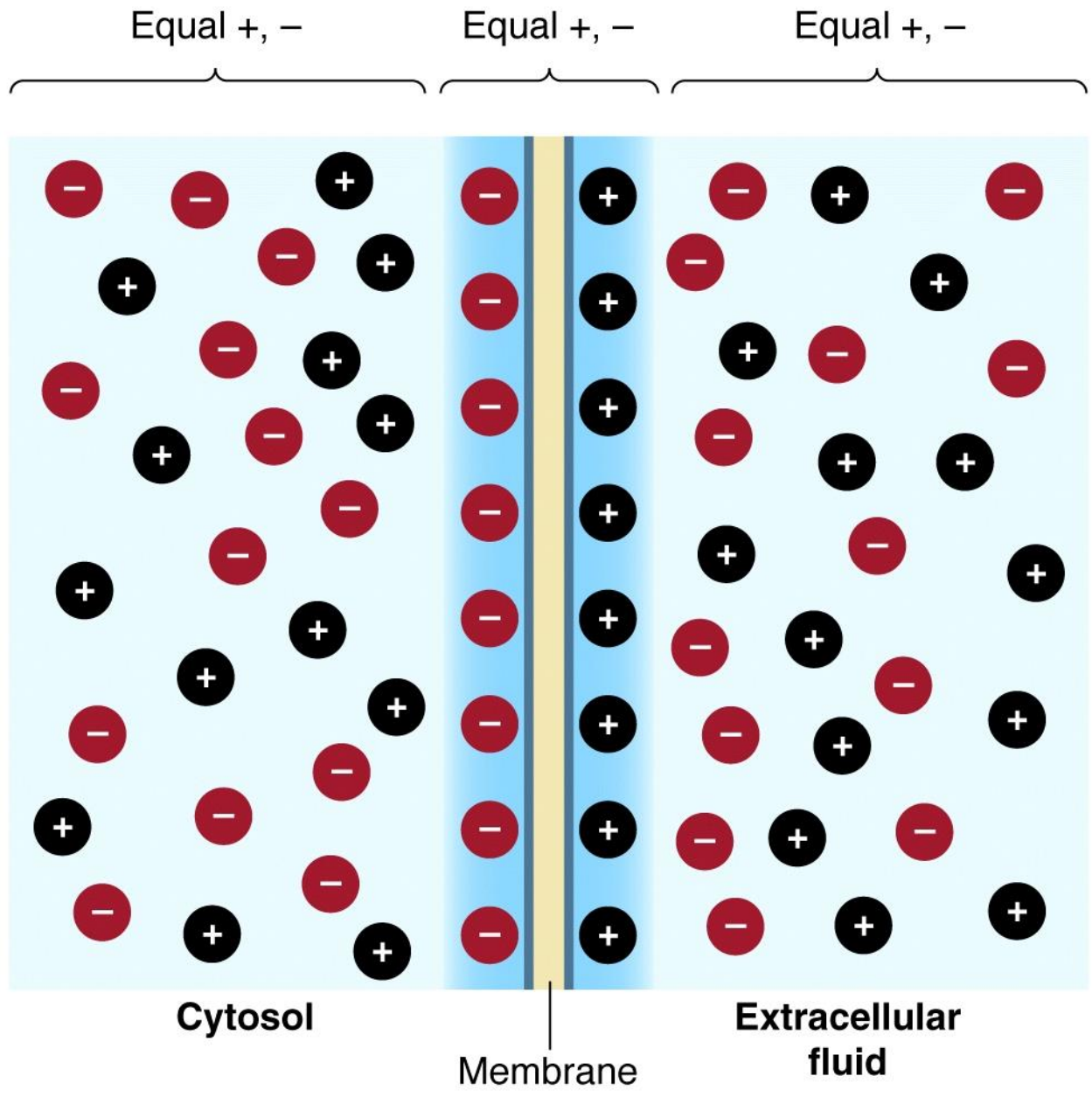


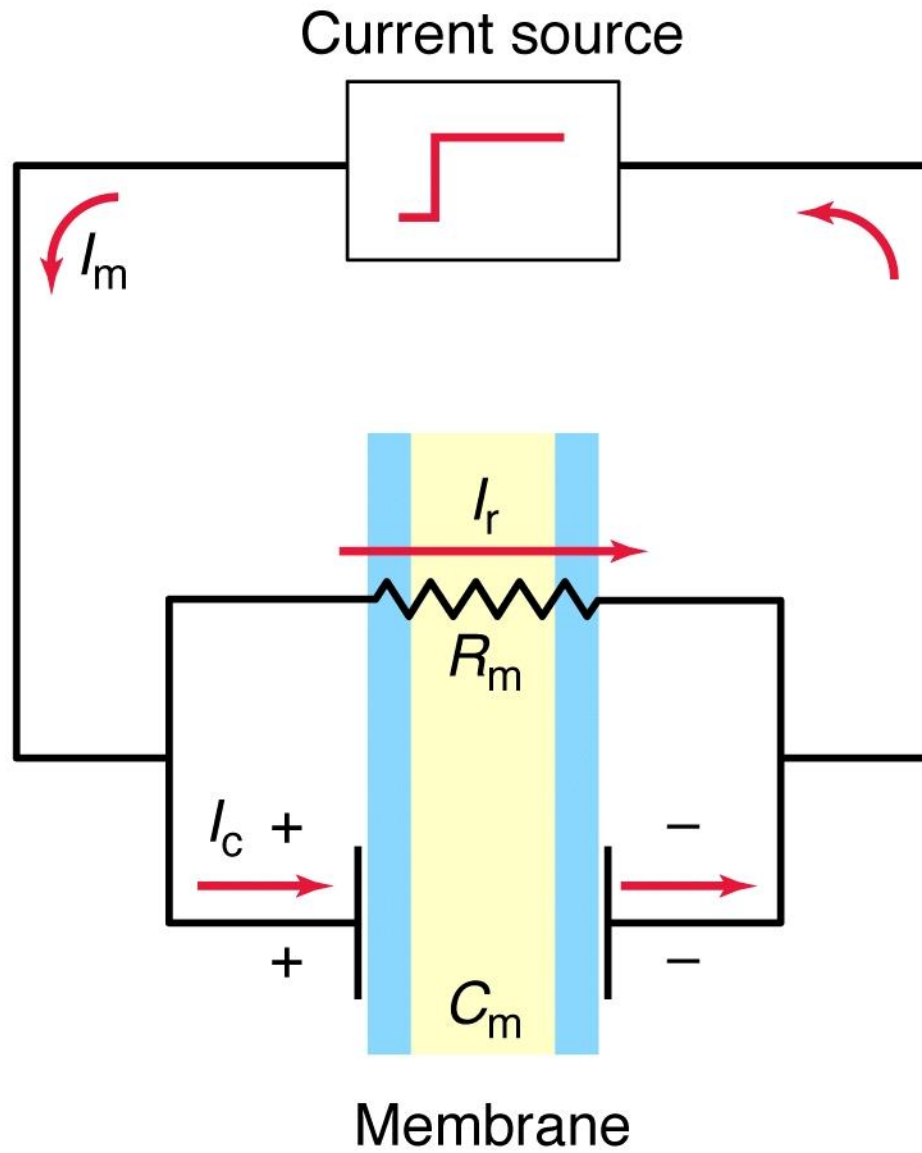
Table 5-1 Examples of ion channels found in axons

Channel	Current through channel	Characteristics	Selected blockers	Function
Leak channel (open in resting axon)	I_K (leak)	Produces relatively high P_K of resting cell	Partially blocked by tetraethylammonium (TEA)	Largely responsible for V_{rest}
Voltage-gated Na^+ channel	I_{Na}	Rapidly activated by depolarization; becomes inactivated even if V_m remains depolarized	Tetrodotoxin (TTX)	Produces rising phase of AP
Voltage-gated Ca^{2+} channel	I_{Ca}	Activated by depolarization but more slowly than Na^+ channel; inactivated as function of cytoplasmic $[Ca^{2+}]$ or V_m	Verapamil, D600, Co^{2+} , Cd^{2+} , Mn^{2+} , Ni^{2+} , La^{3+}	Produces slow depolarization; allows Ca^{2+} to enter cell, where it can act as second messenger
Voltage-gated K^+ channel (“delayed rectifier”)	$I_{K(V)}$	Activated by depolarization but more slowly than Na^+ channel; inactivated slowly and not completely if V_m remains depolarized	Intra- and extracellular TEA, amino pyridines	Carries current that rapidly repolarizes the membrane to terminate an AP
Ca^{2+} -dependent K^+ channel	$I_{K(Ca)}$	Activated by depolarization plus elevated cytoplasmic $[Ca^{2+}]$; remains open as long as cytoplasmic $[Ca^{2+}]$ is higher than normal	Extracellular TEA	Carries current that repolarizes the cell following APs based on either Na^+ or Ca^{2+} and that balances I_{Ca} , thus limiting depolarization by I_{Ca}

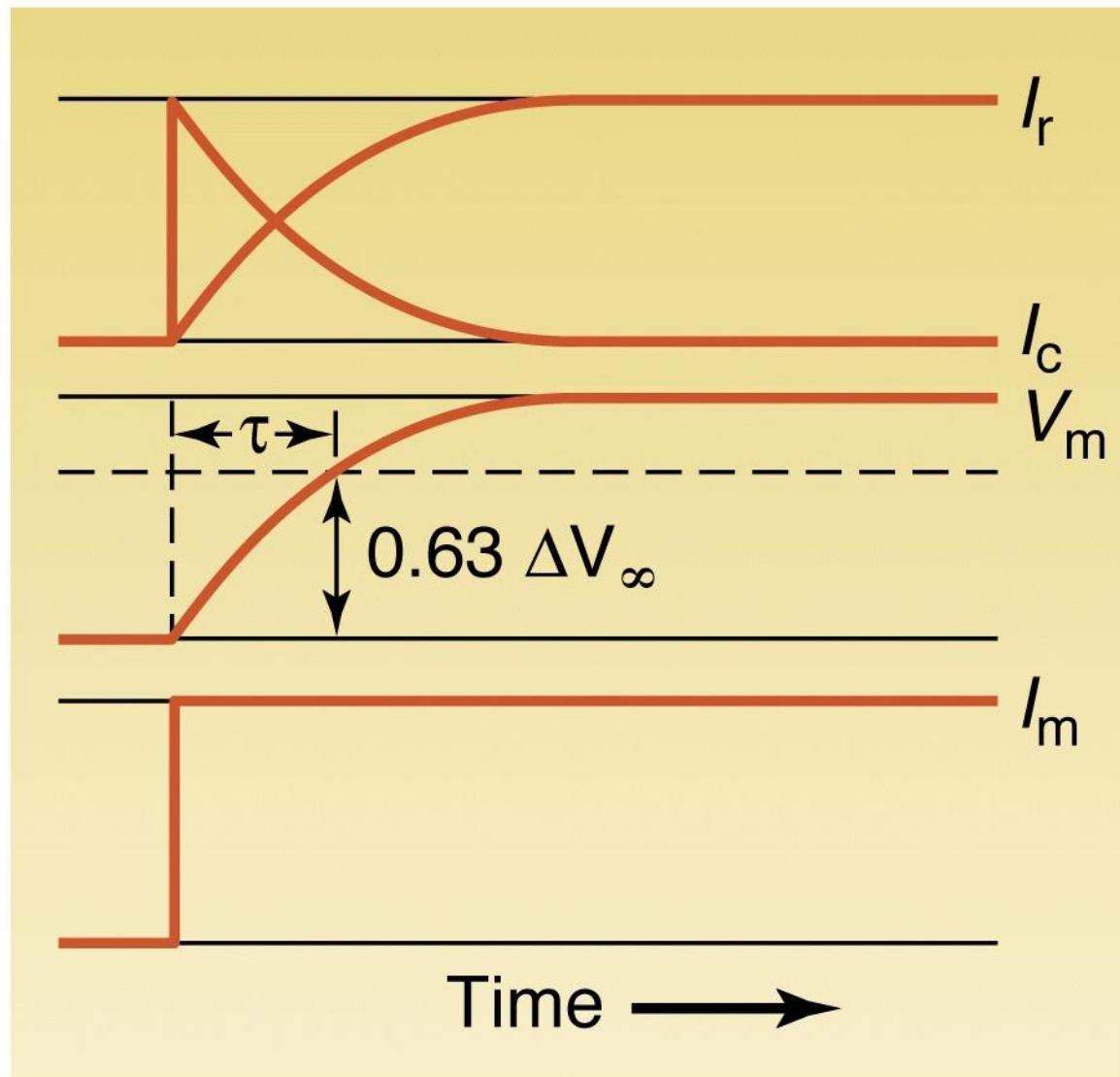




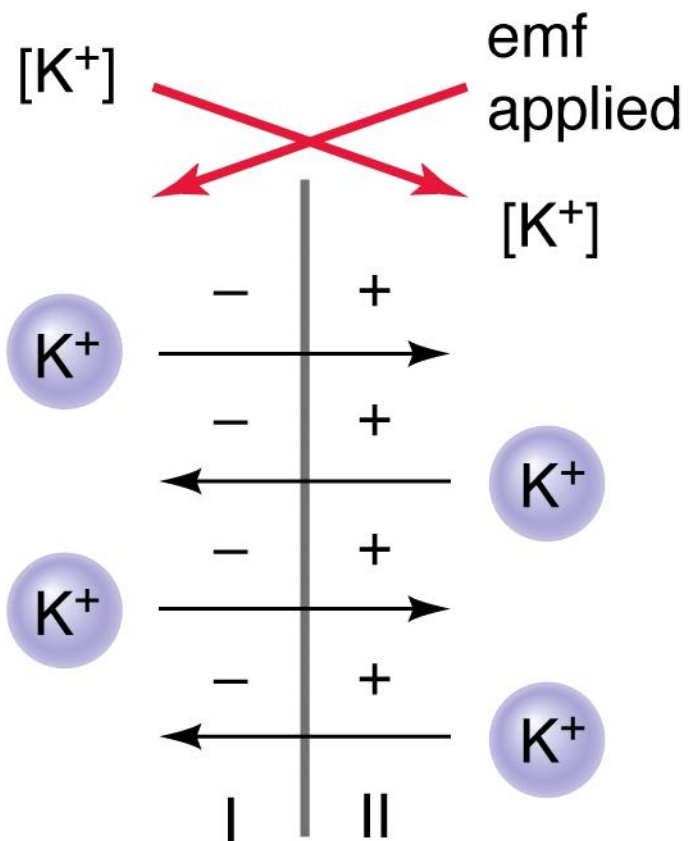
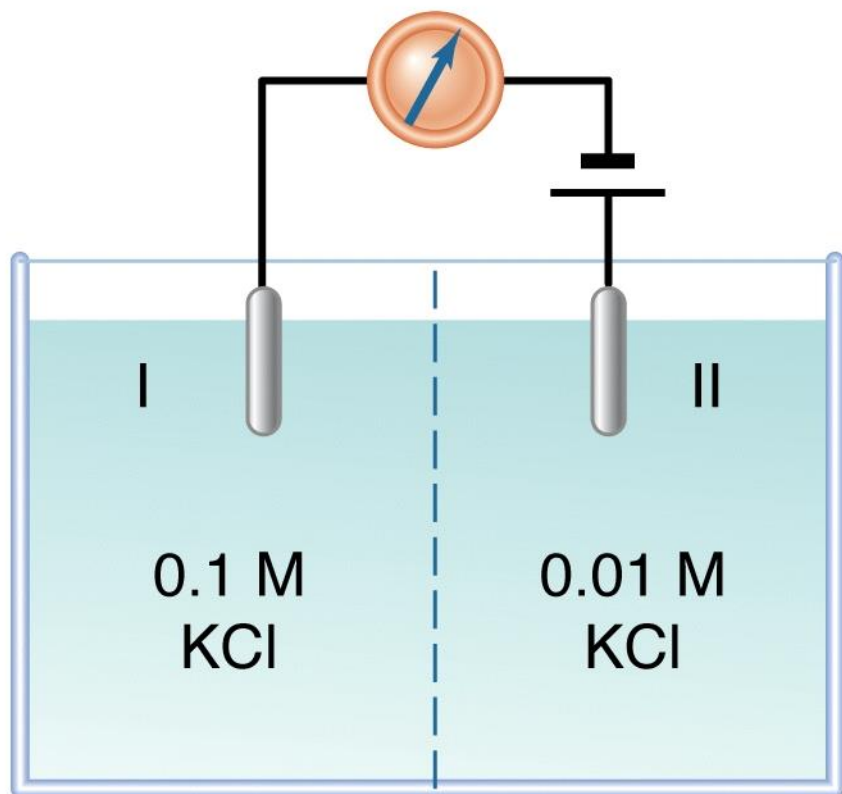
(a)



(b)



(c)



The Nernst equation is generally considered for ions across a membrane generating an electromotive force as commonly shown as:

$$V = \frac{RT}{zF} \cdot \ln \frac{[X]_{out}}{[X]_{in}}$$

X = ion of interest

V = equilibrium voltage for the X ion across the membrane

R = gas constant [8.314 J/(mol•K)]

T = absolute temperature [Kelvin]

Z = valence of the ion

F = Faraday's constant [9.649 × 10⁴ C/mol]

For the K⁺ ion at 20°C and transformation of ln to log₁₀ along with filling in the constants, one arrives at:

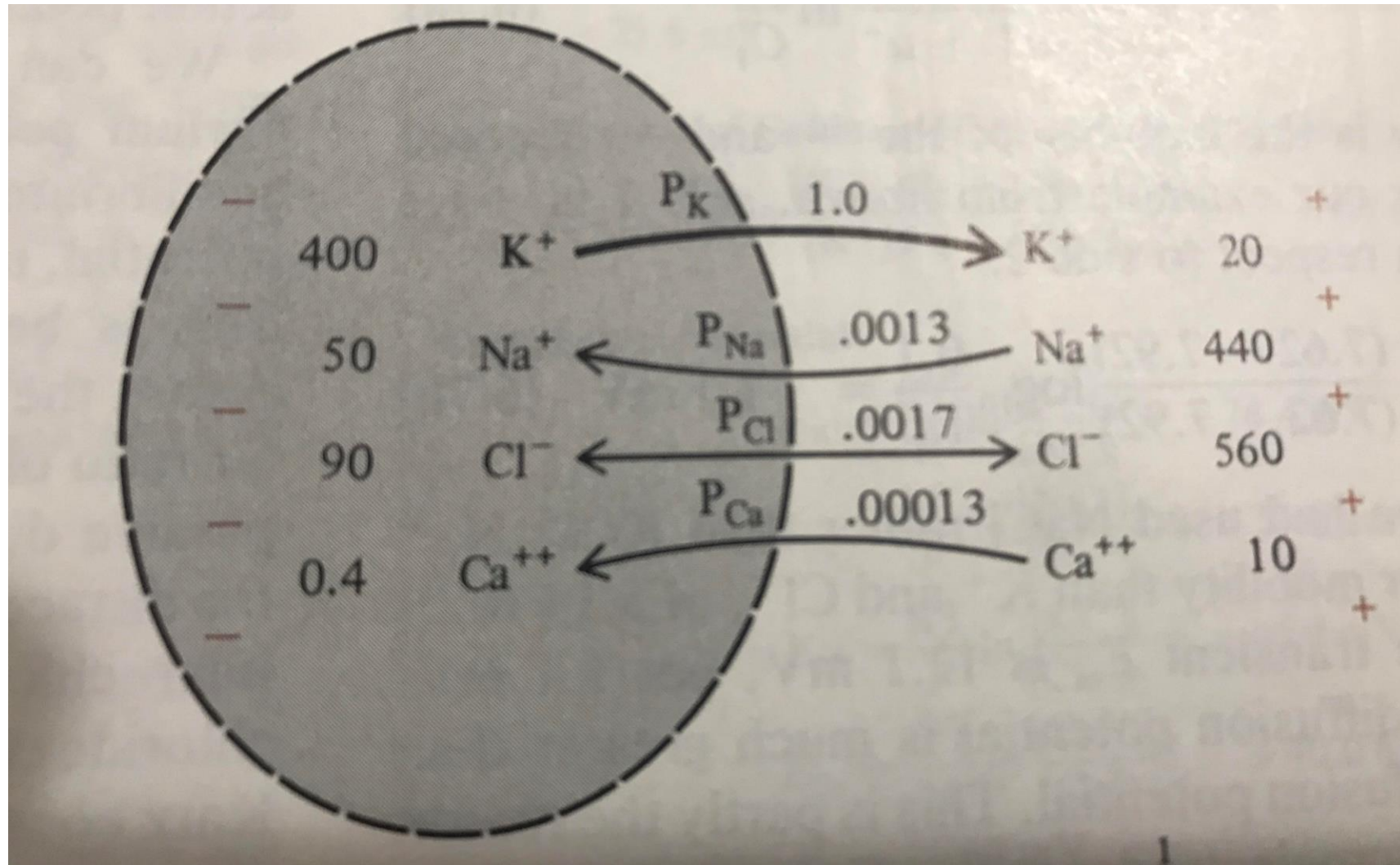
$$Potential = 58 \log \frac{[K]_{out}}{[K]_{in}}$$

Here is a generalized G-H-K equation for Na⁺, K⁺, and Cl⁻ ions:

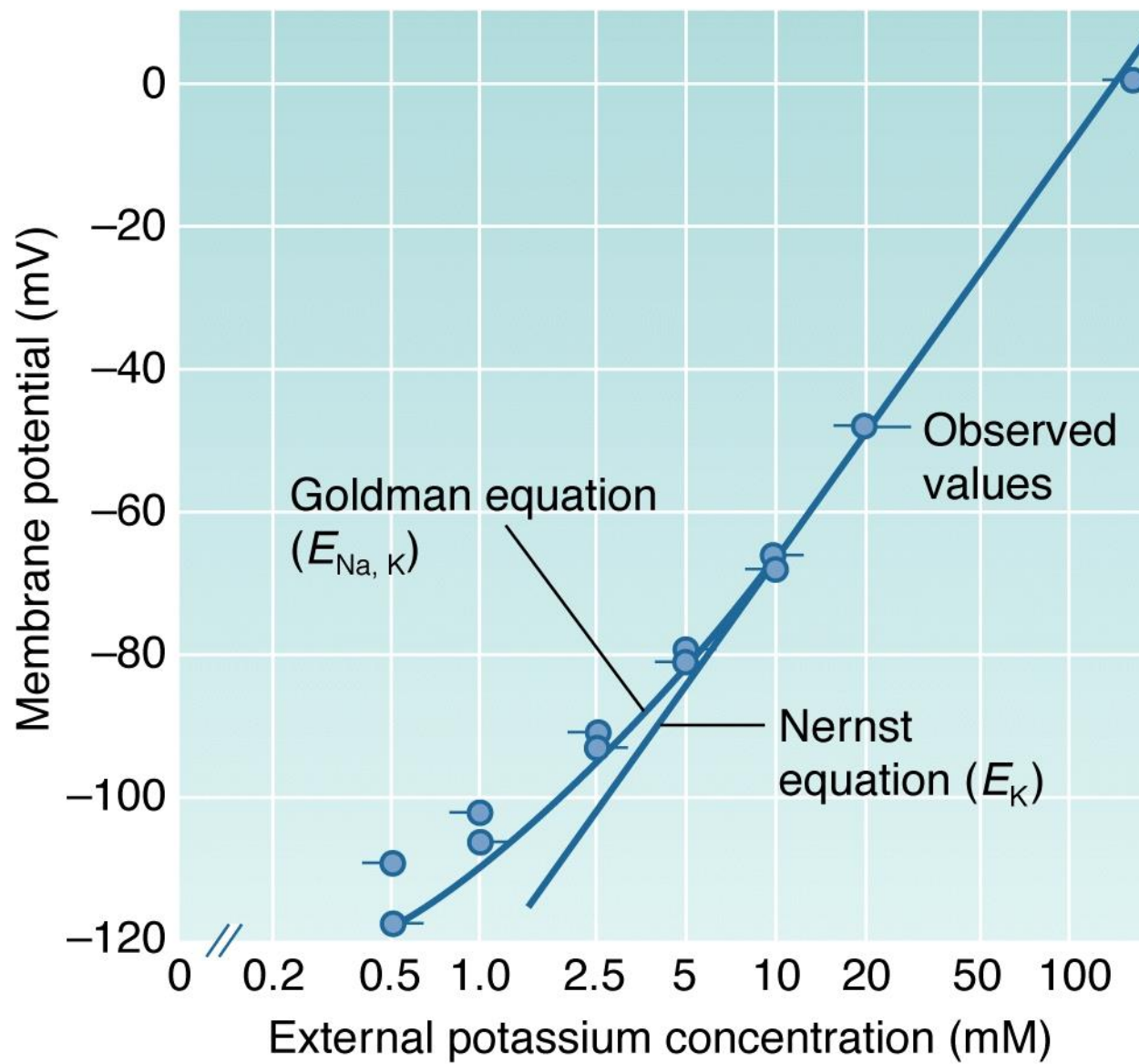
Goldman-Hodgkin-Katz (G-H-K) equation,

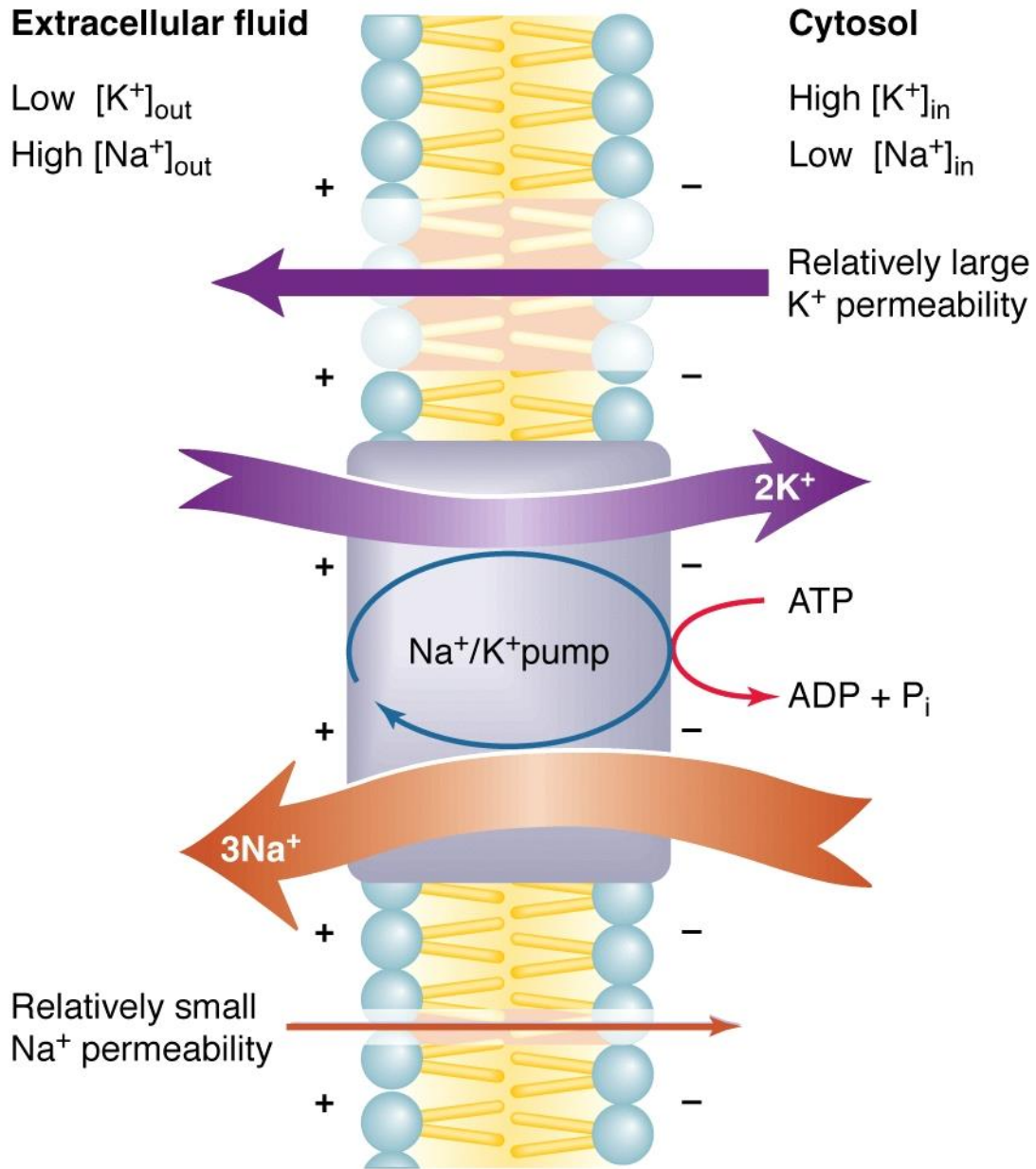
$$Em_{K,Na,Cl} = \frac{RT}{F} \ln \frac{P_{Na^+}[Na^+]_{out} + P_{K^+}[K^+]_{out} + P_{Cl^-}[Cl^-]_{in}}{P_{Na^+}[Na^+]_{in} + P_{K^+}[K^+]_{in} + P_{Cl^-}[Cl^-]_{out}}$$

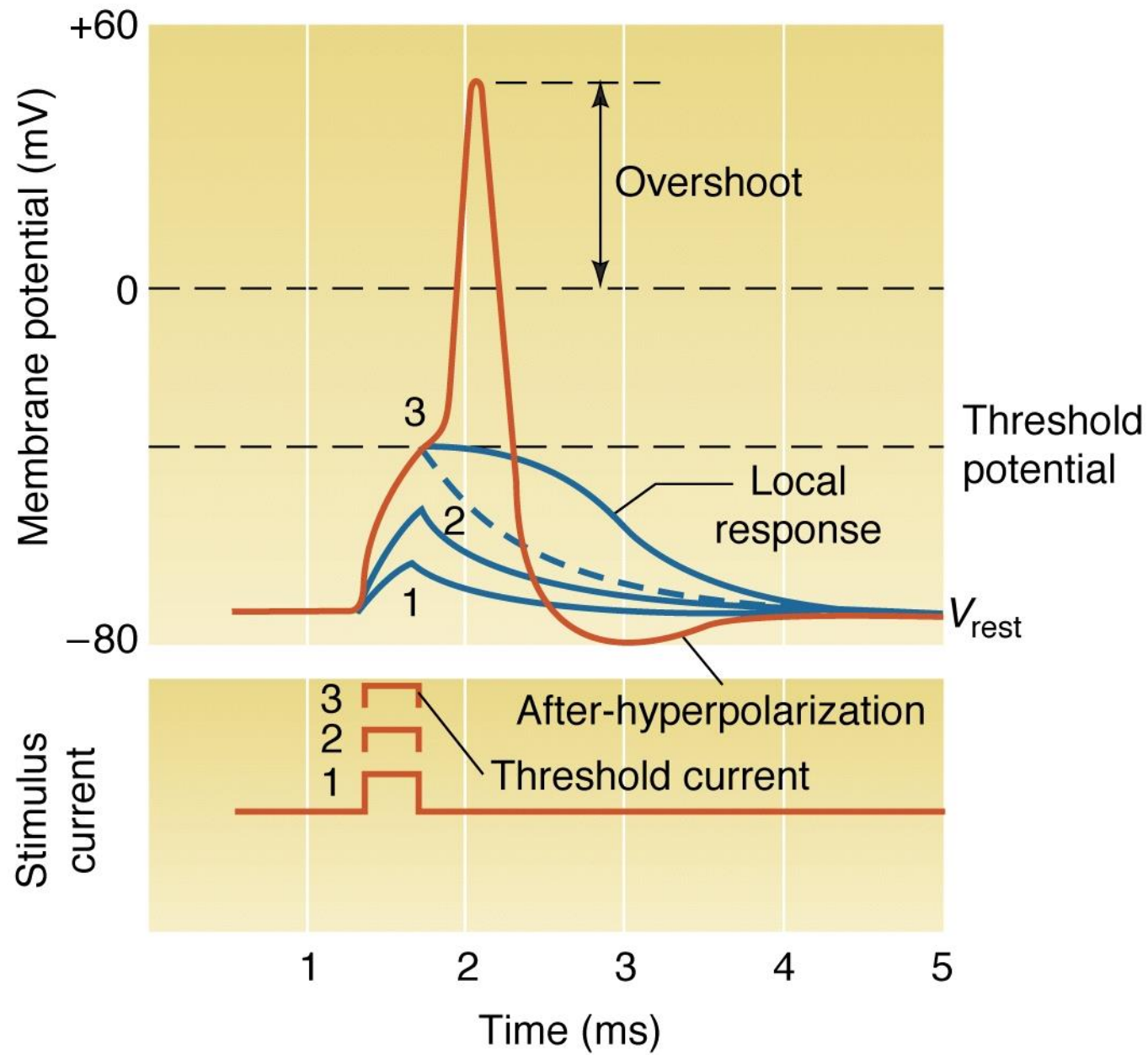
Squid neuron

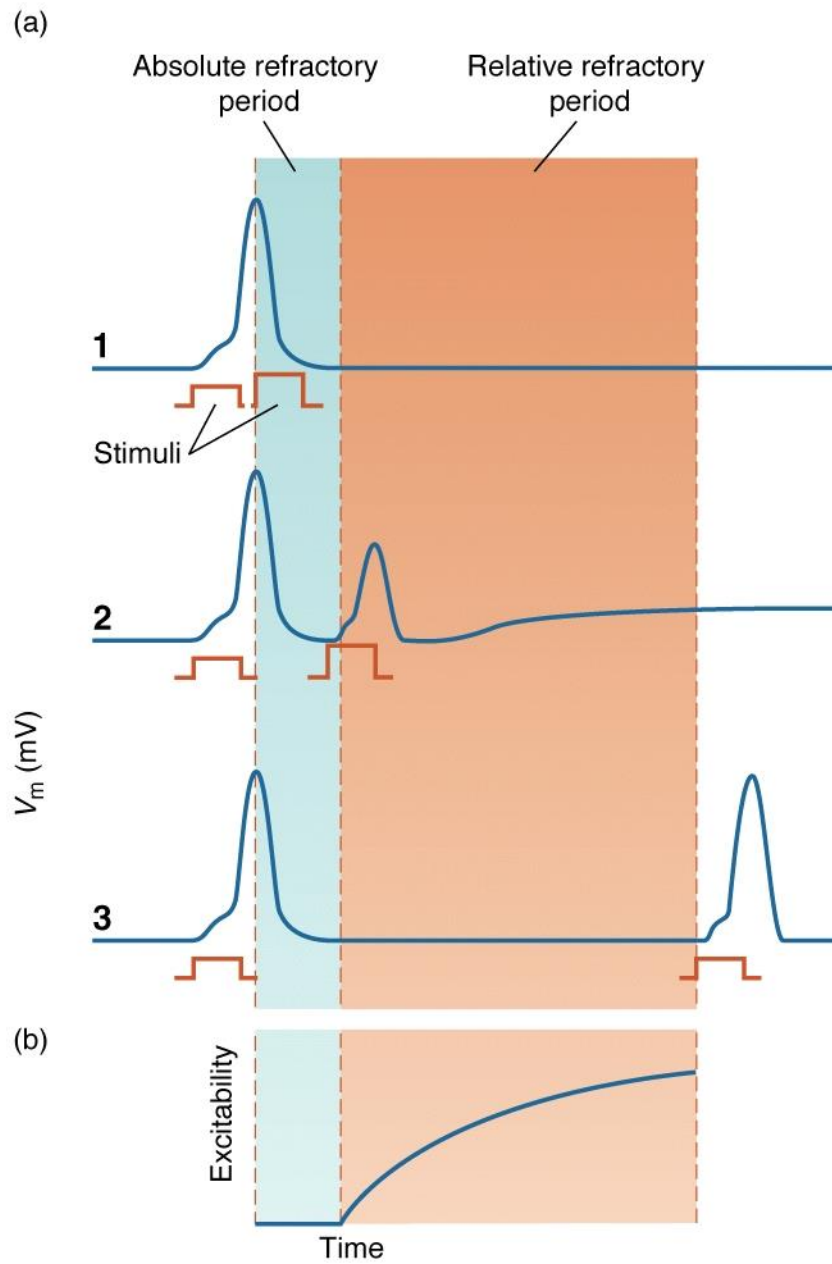


(b)

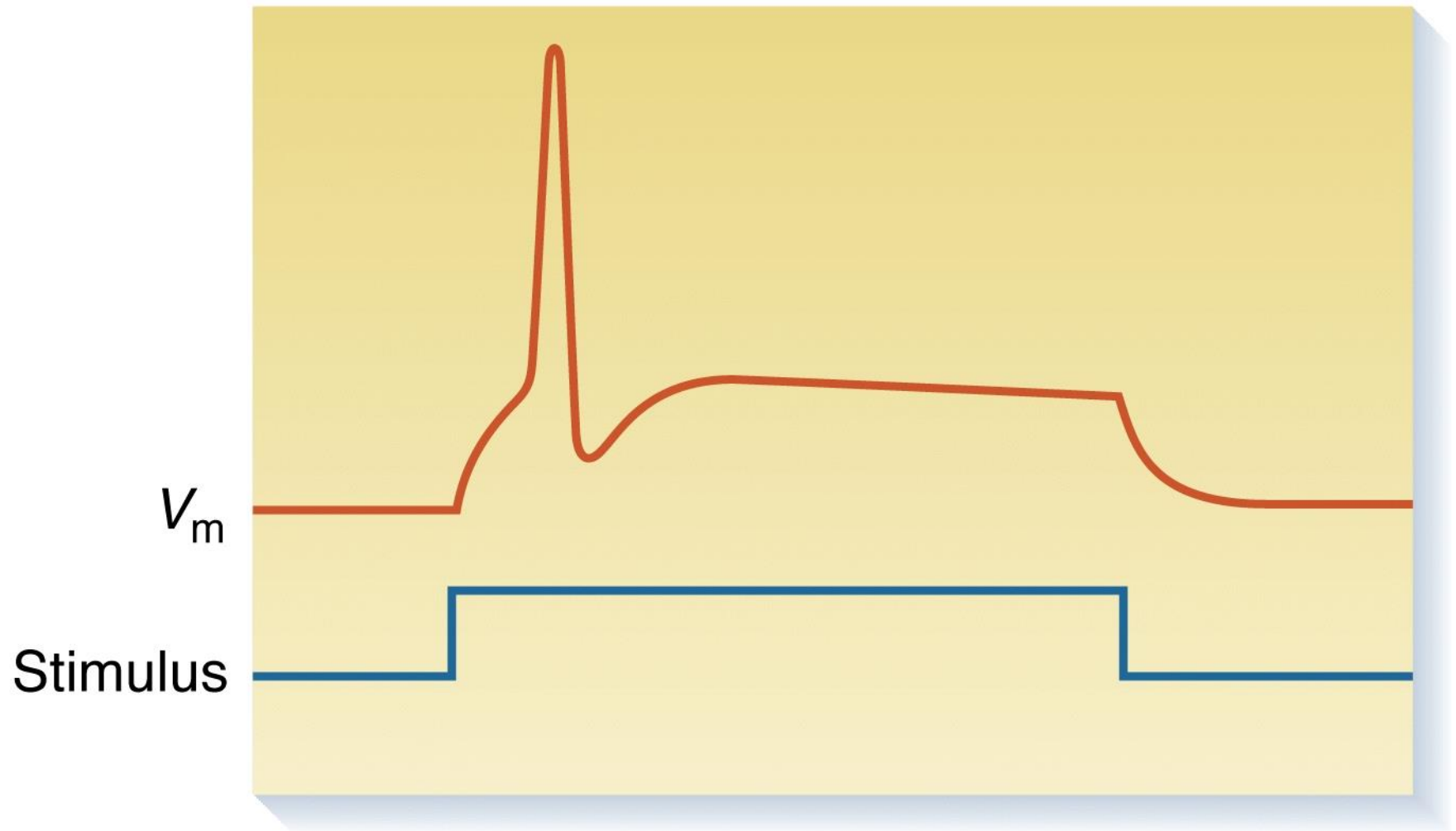




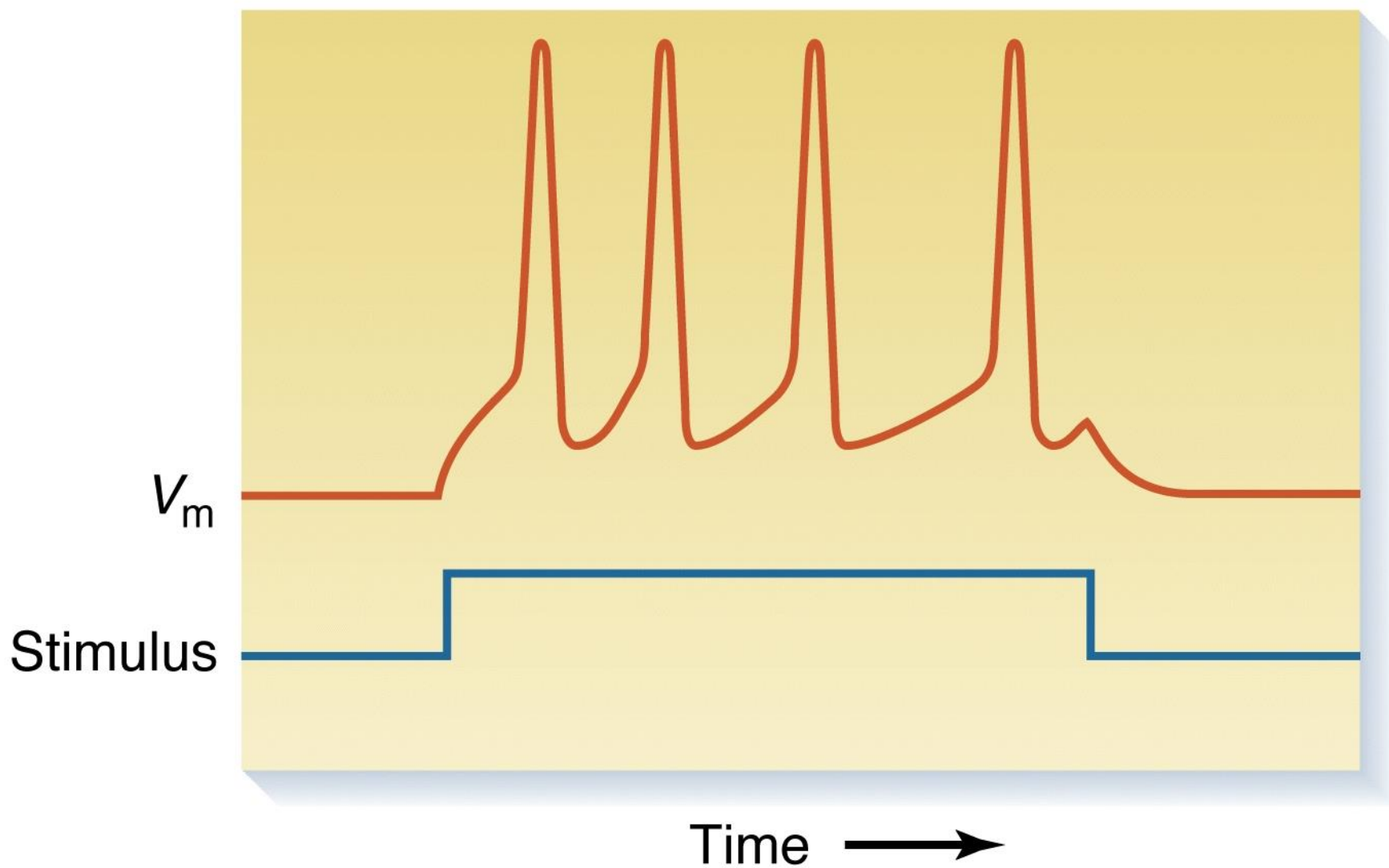




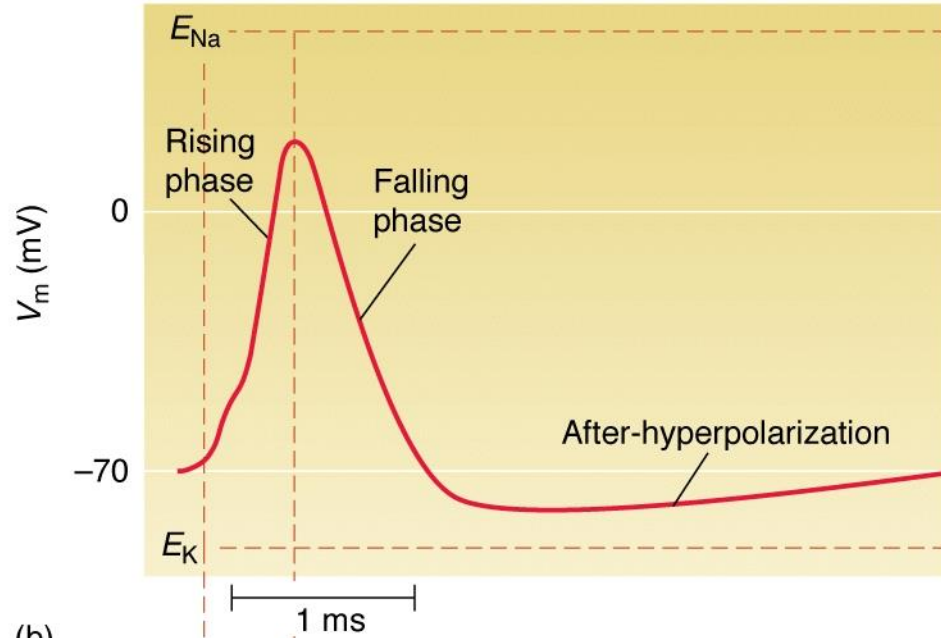
(a) Phasic response



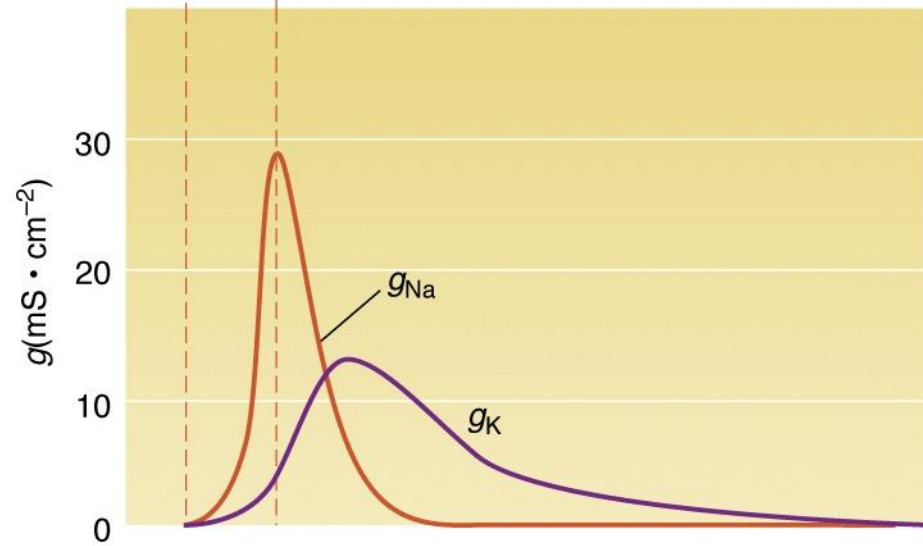
(b) Tonic response



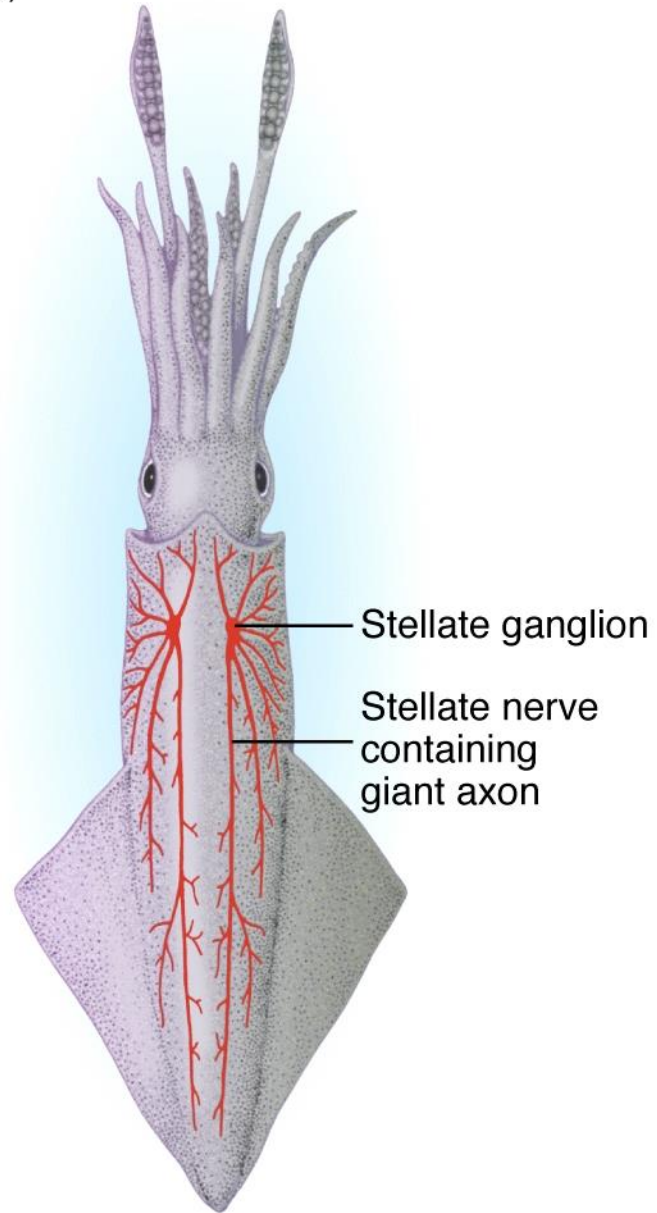
(a)



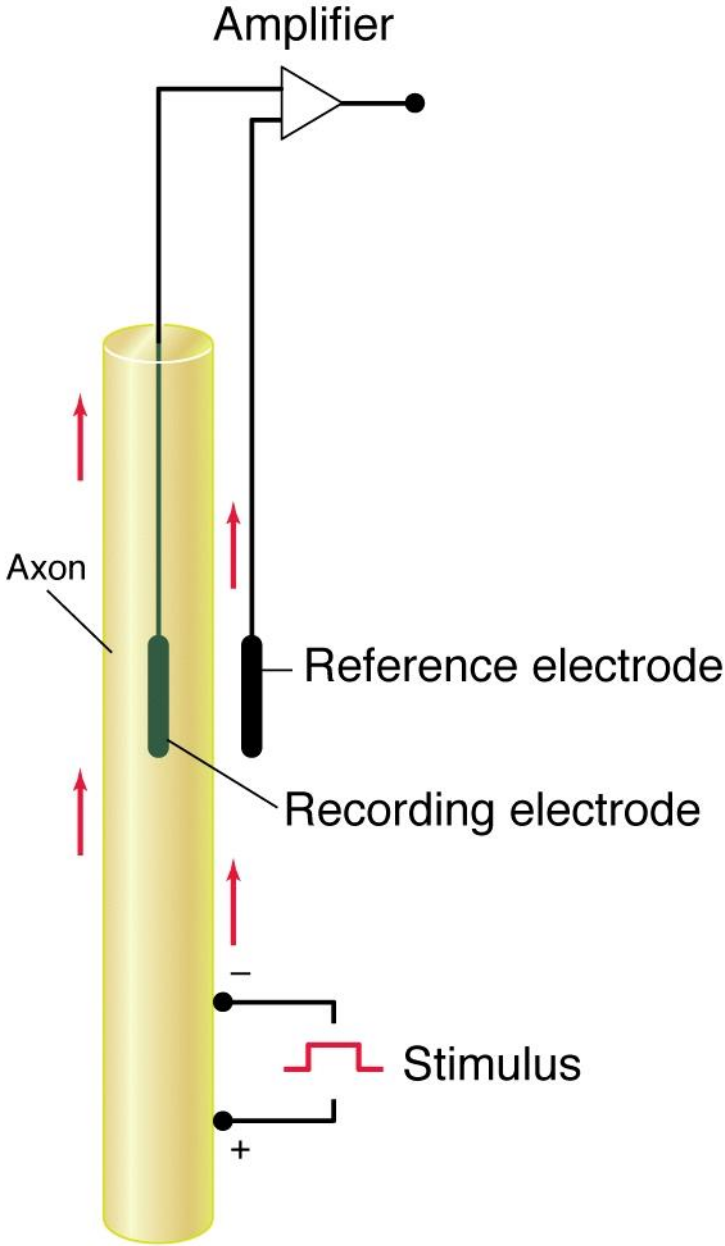
(b)



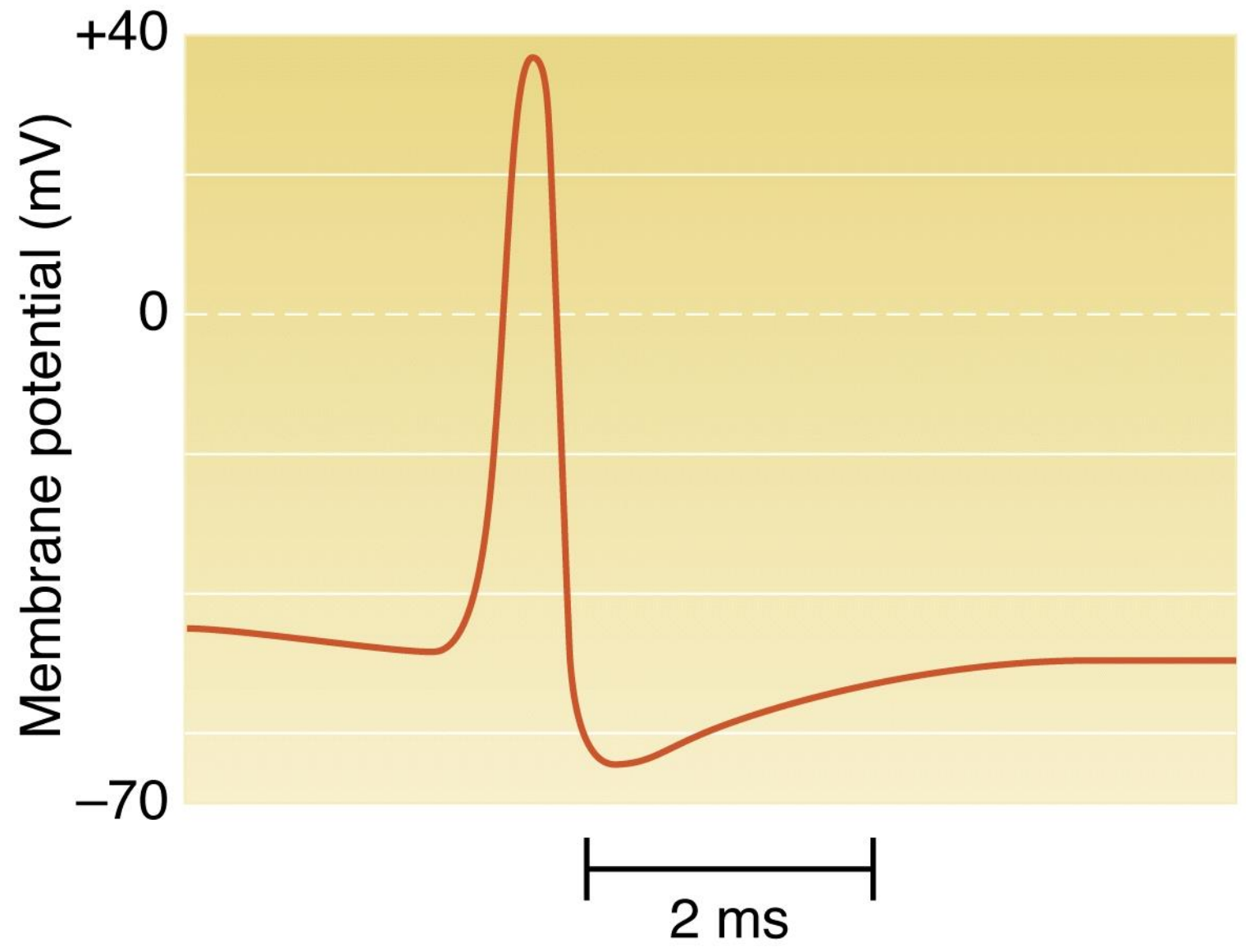
(a)



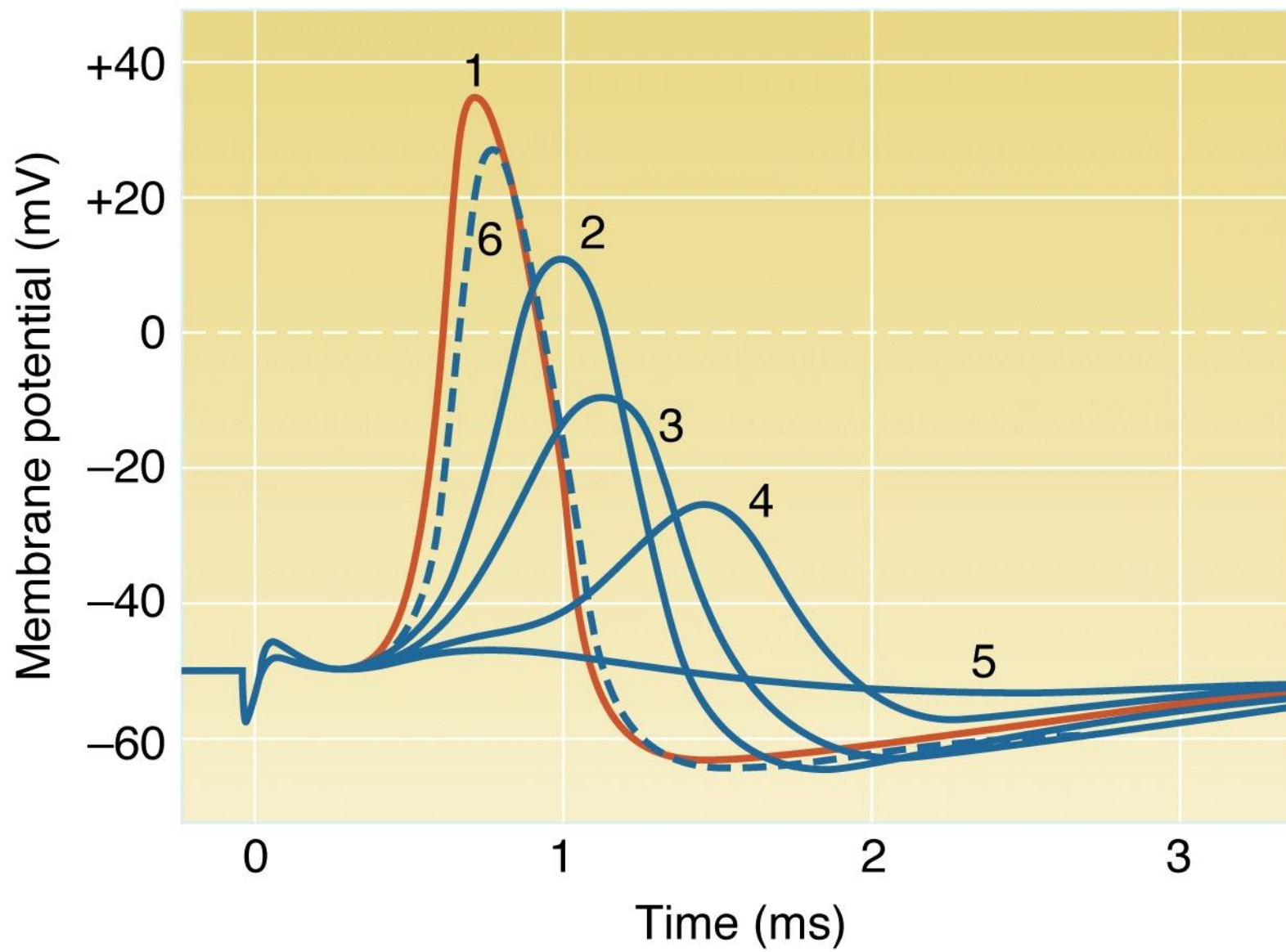
(b)

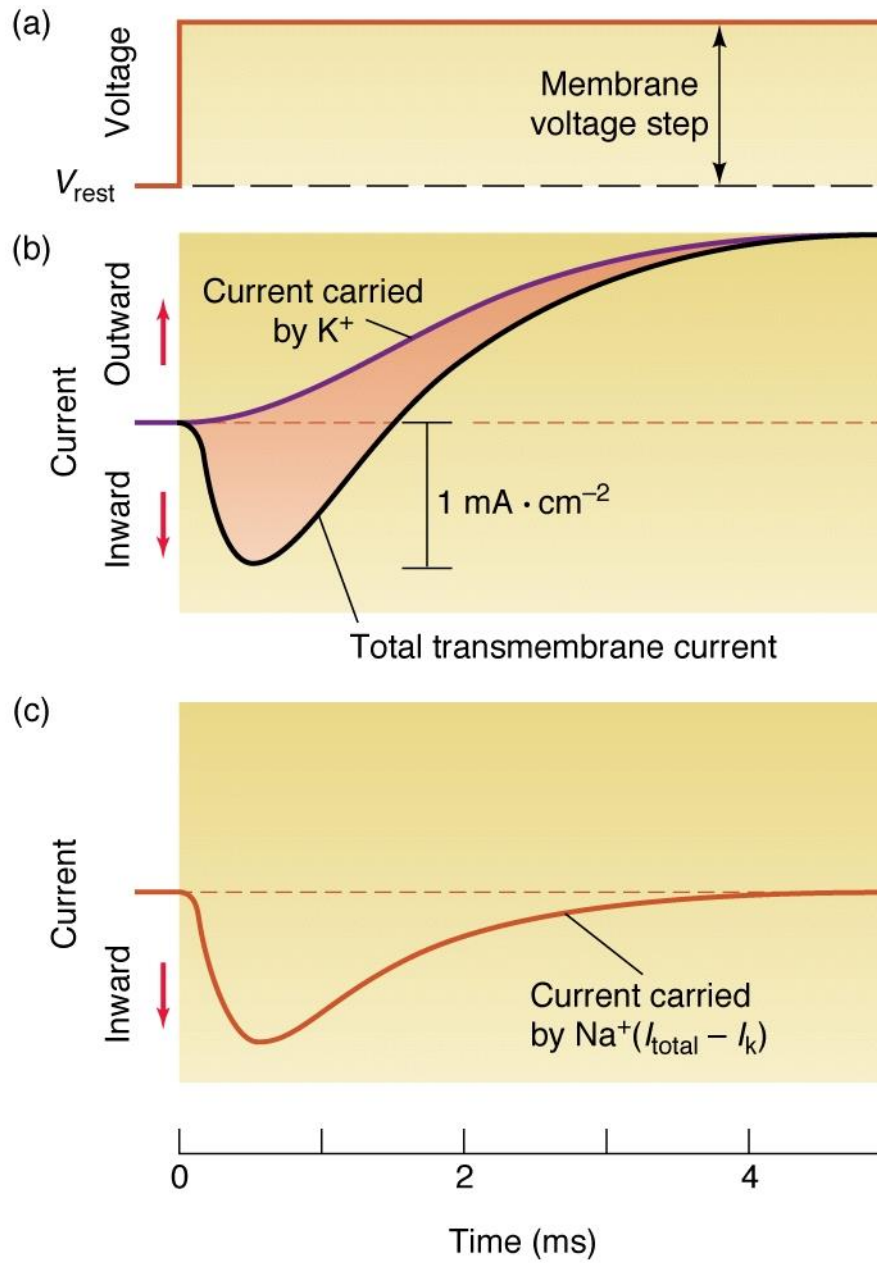


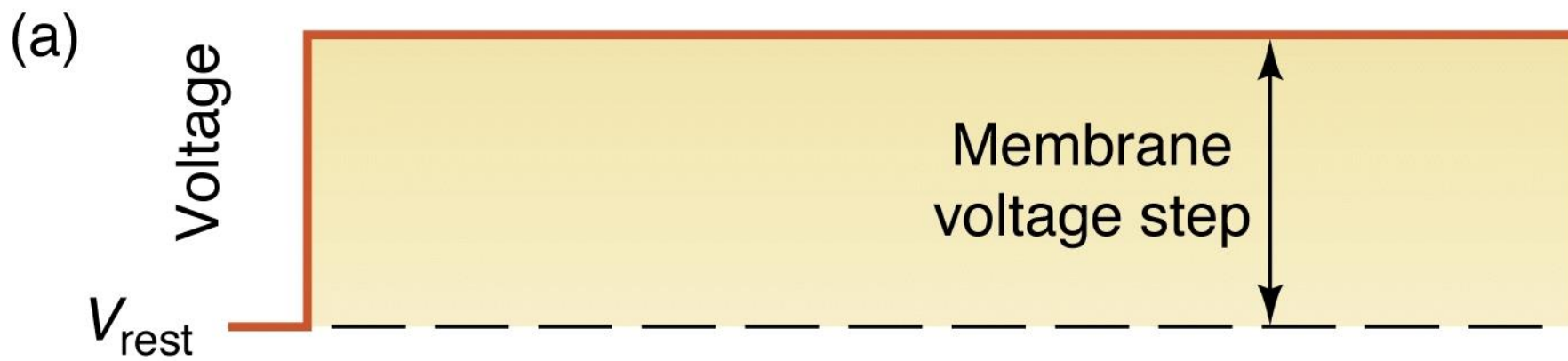
(c)



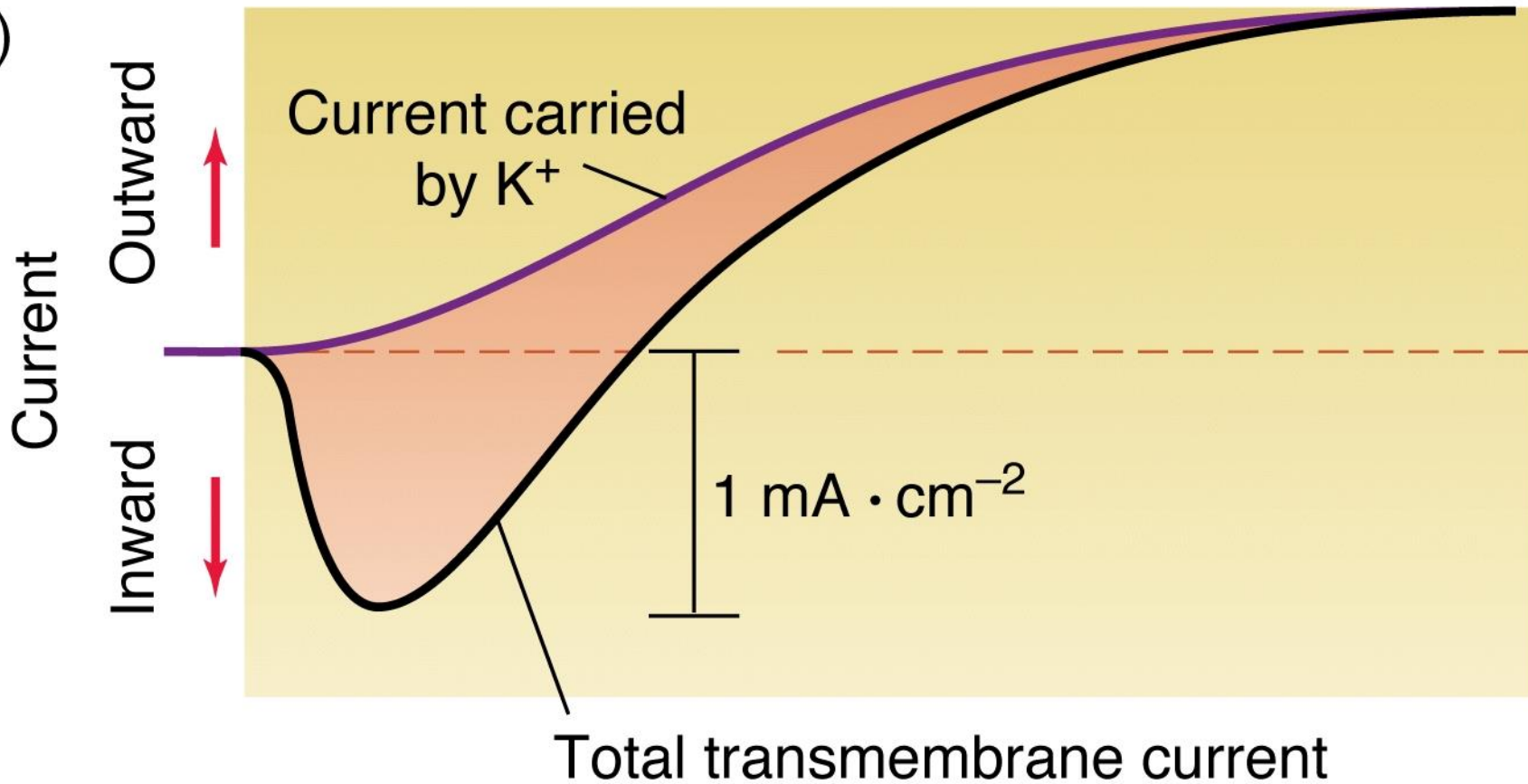
(d)

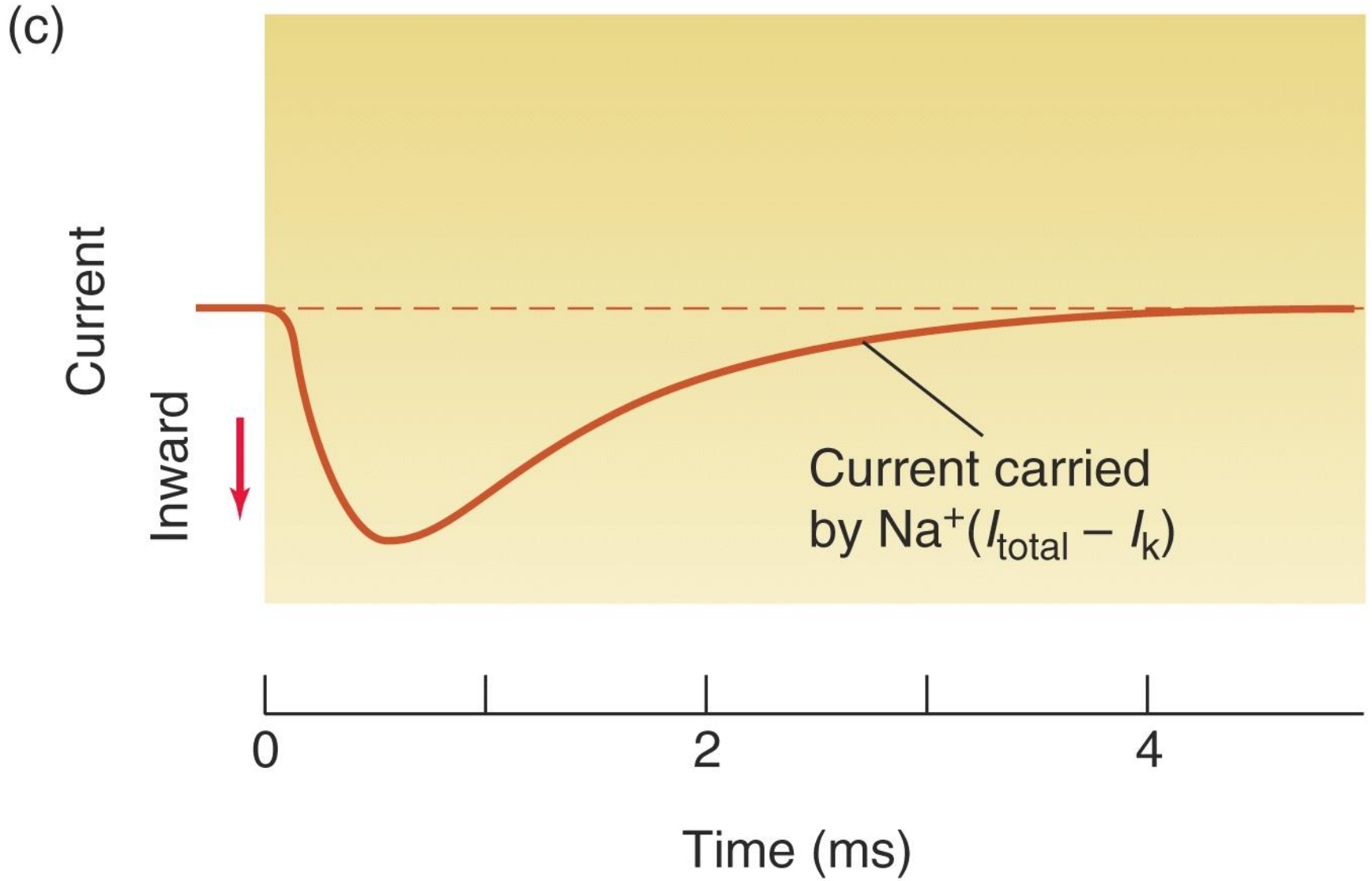




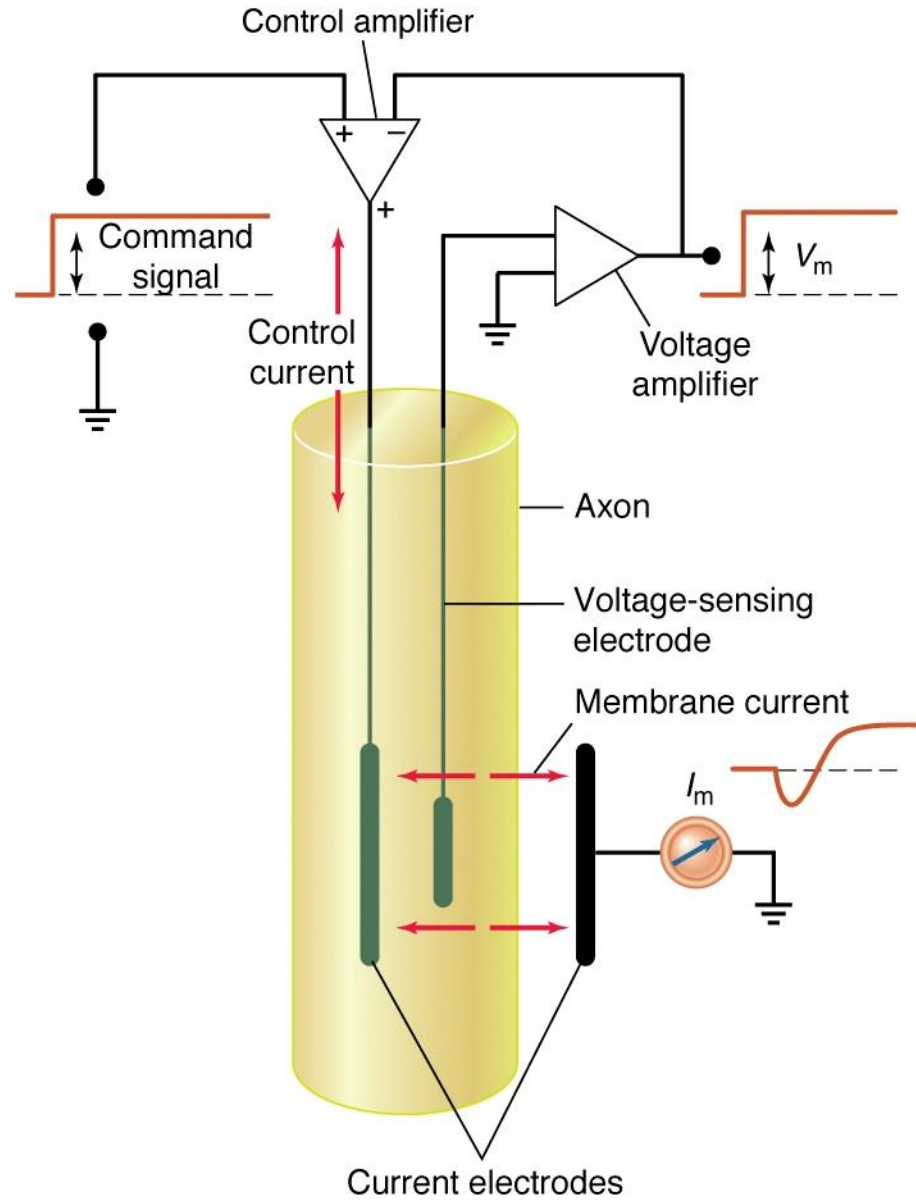


(b)



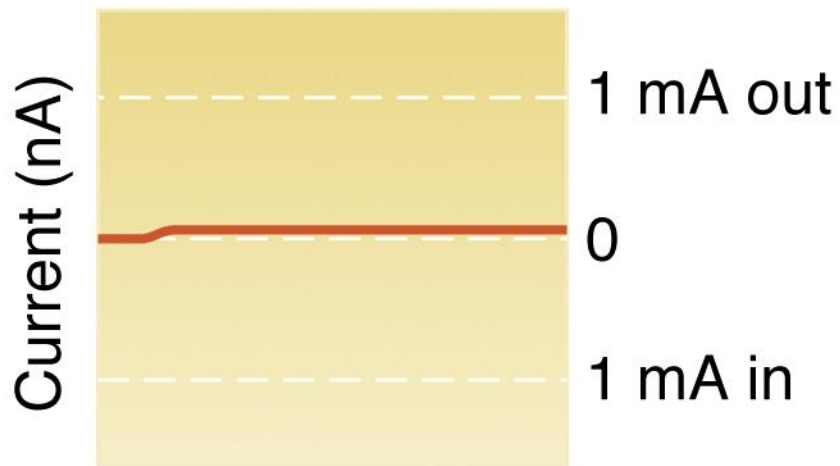


(a)

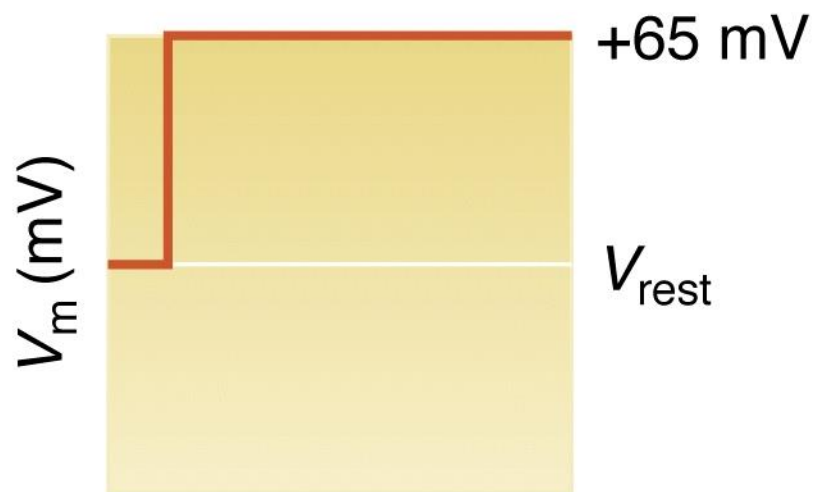
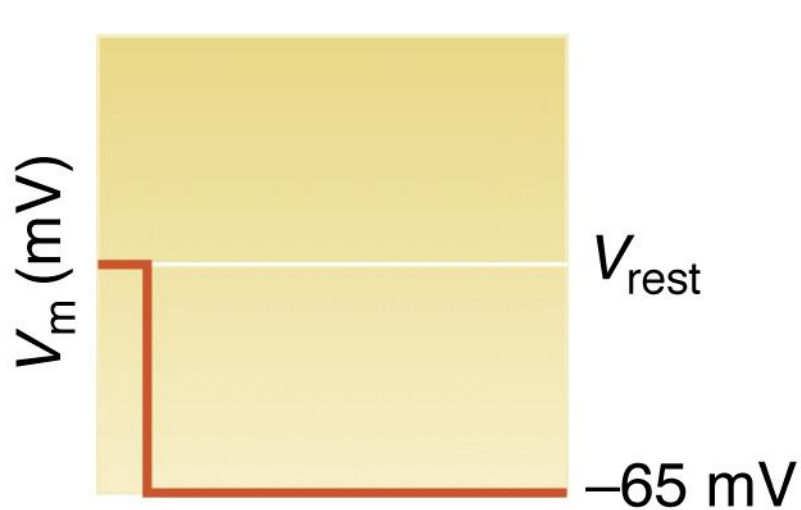
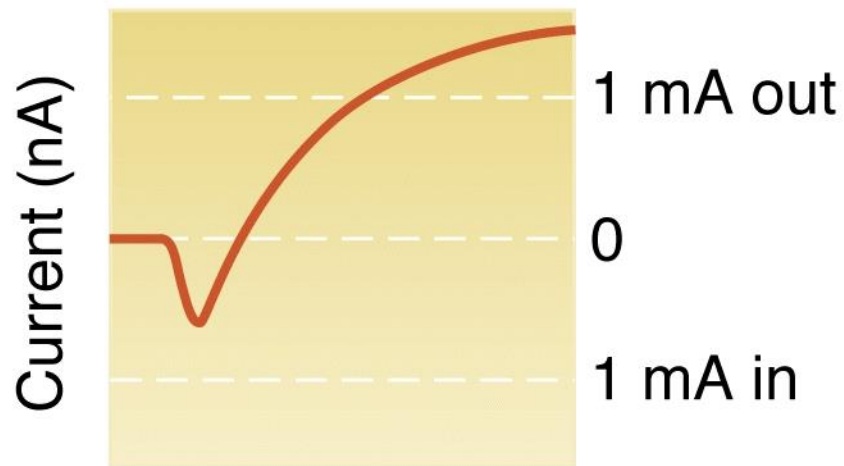


(b)

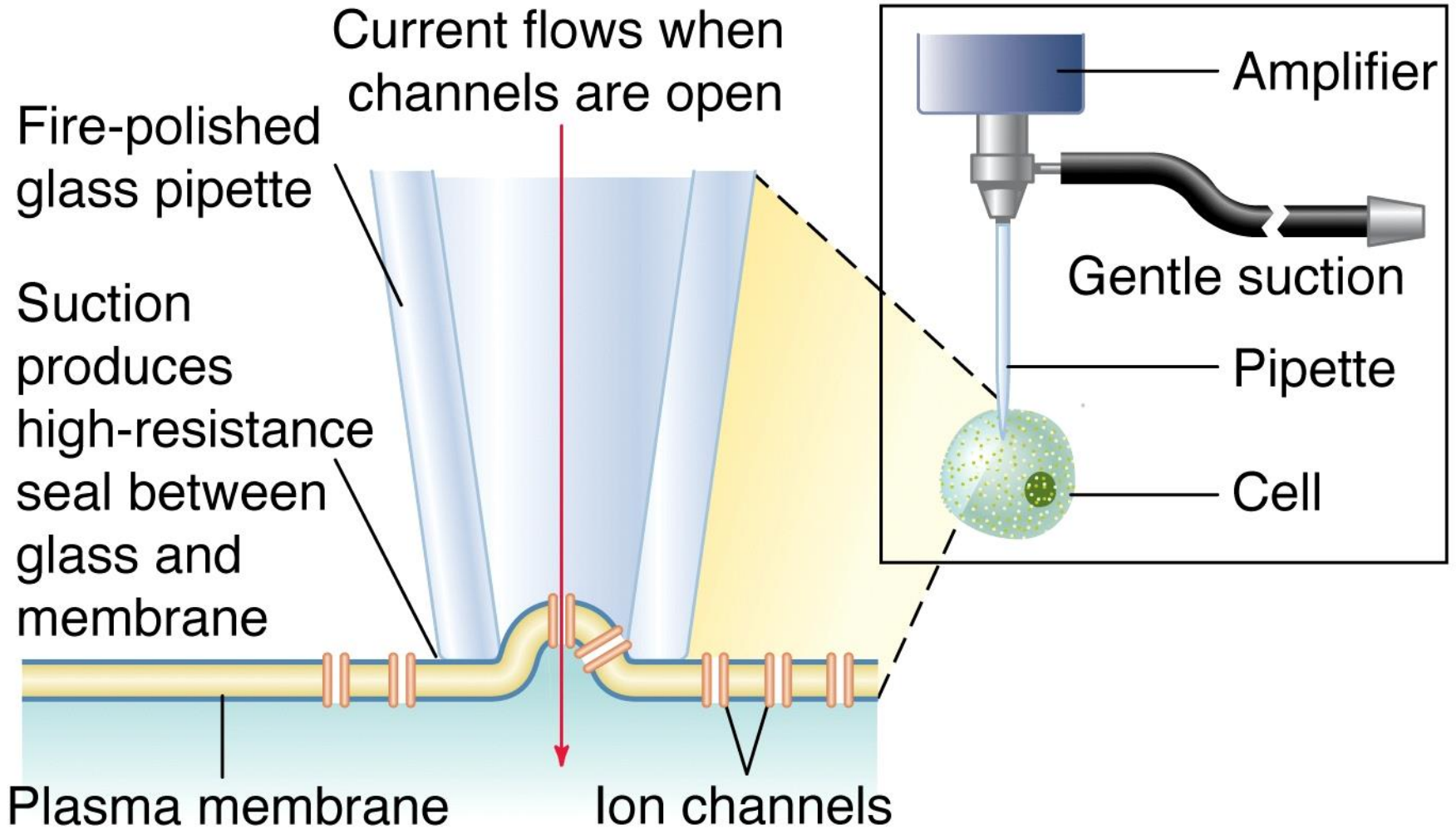
Hyperpolarization



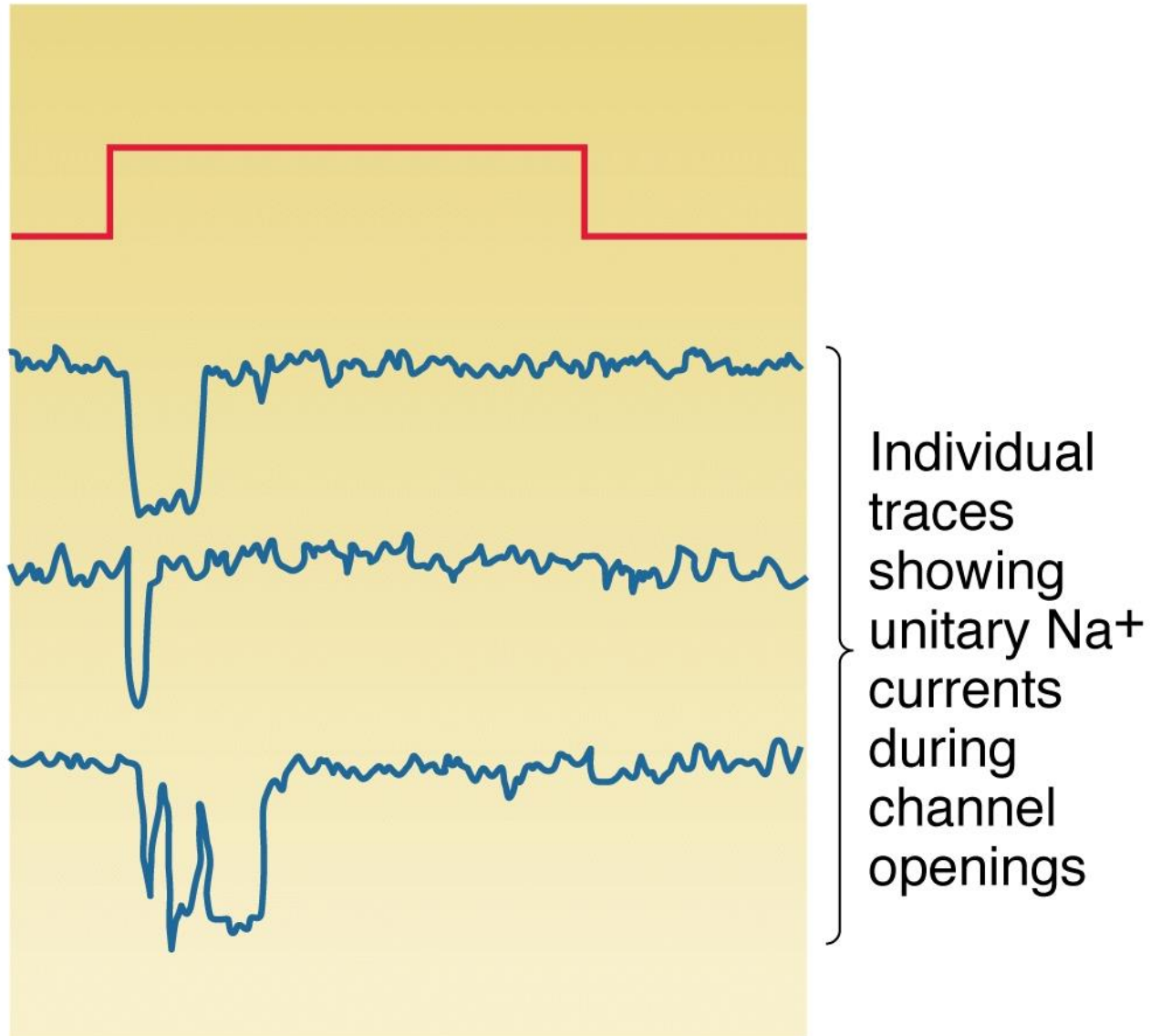
Depolarization



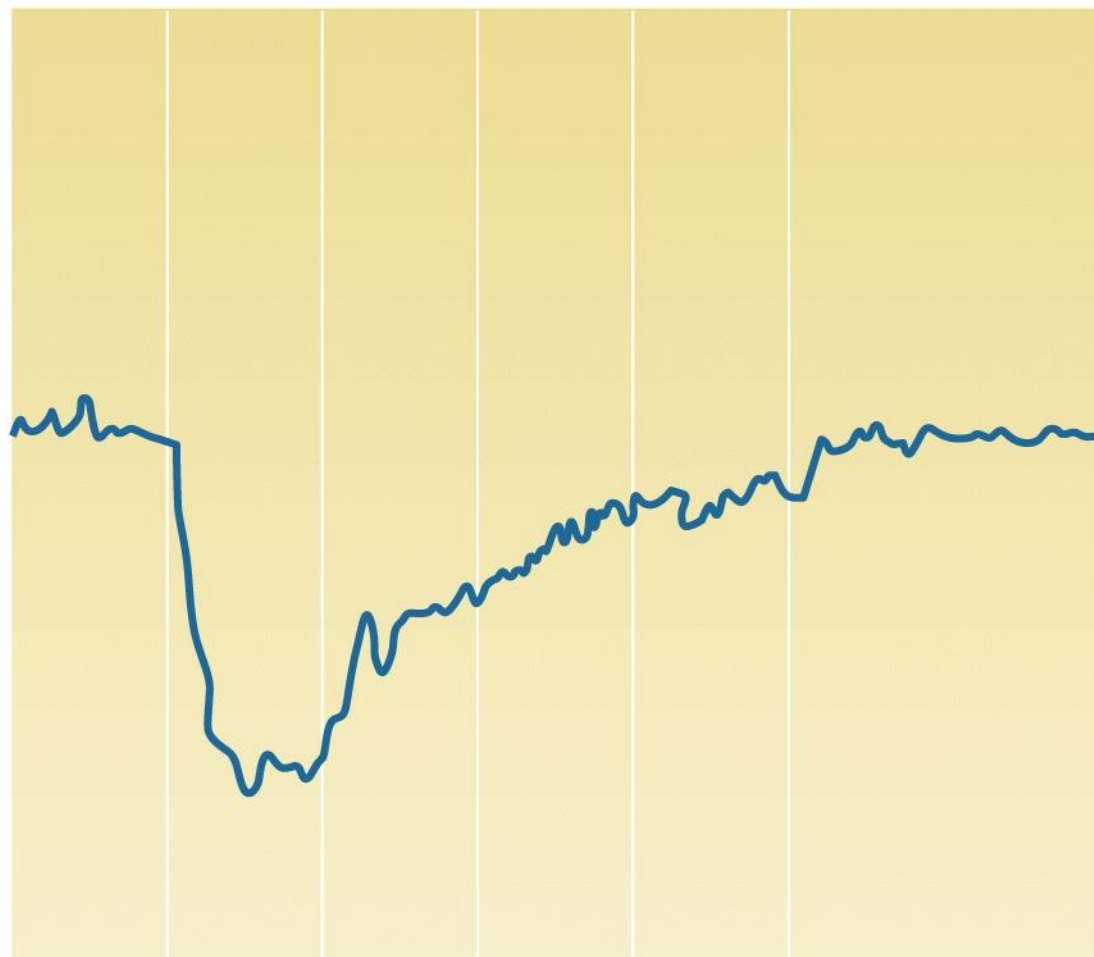
(a)



(b)



(c)



Ensemble
current
reconstructed
by summing
many traces
like those
in part b

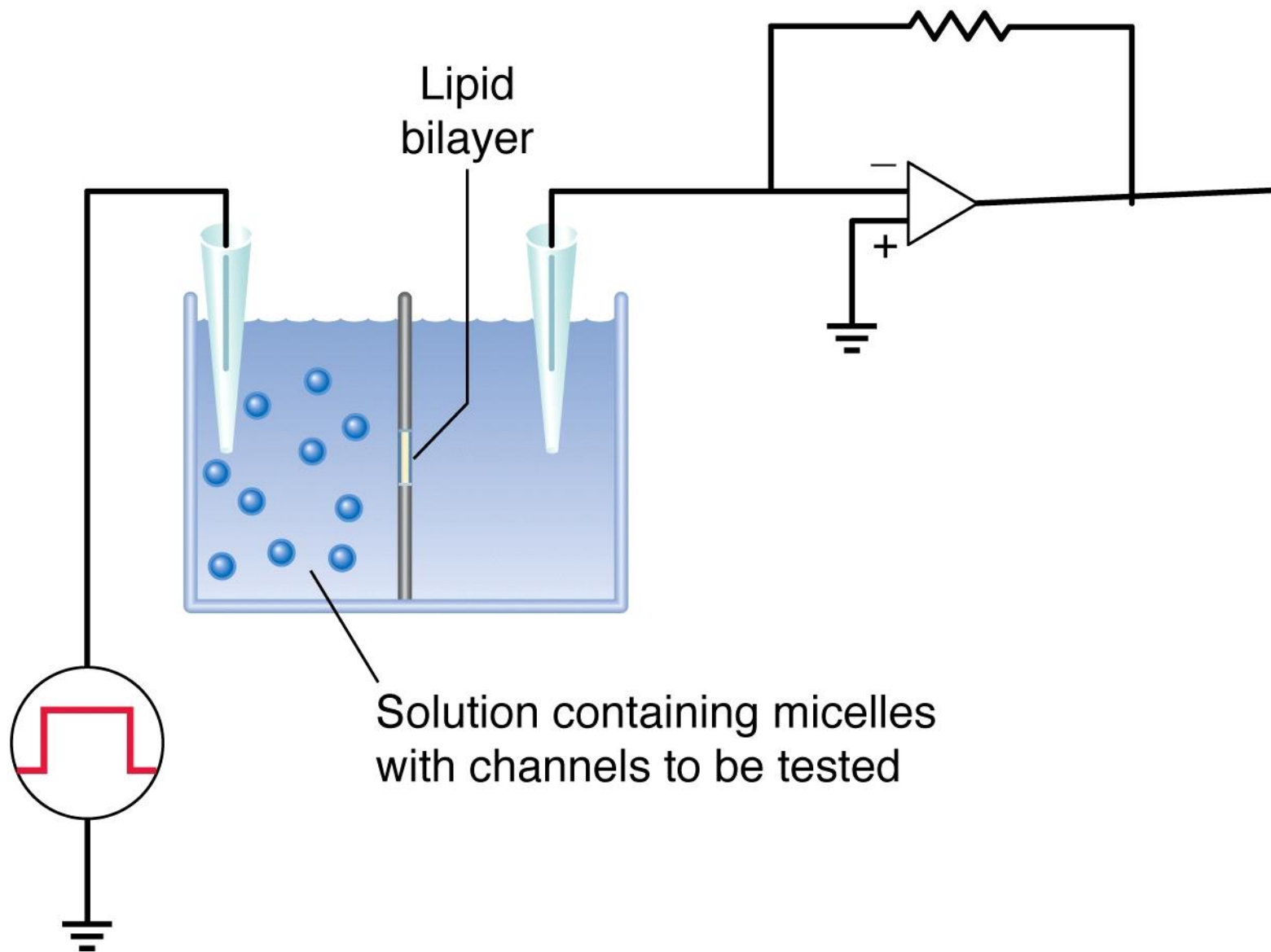
0

30

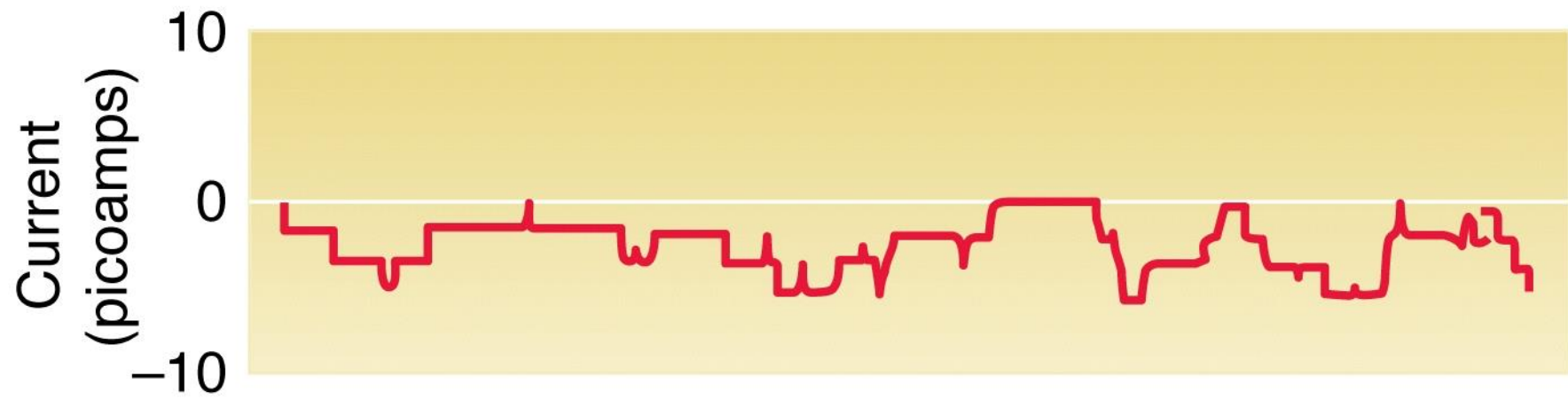
60

Time (ms)

(a)

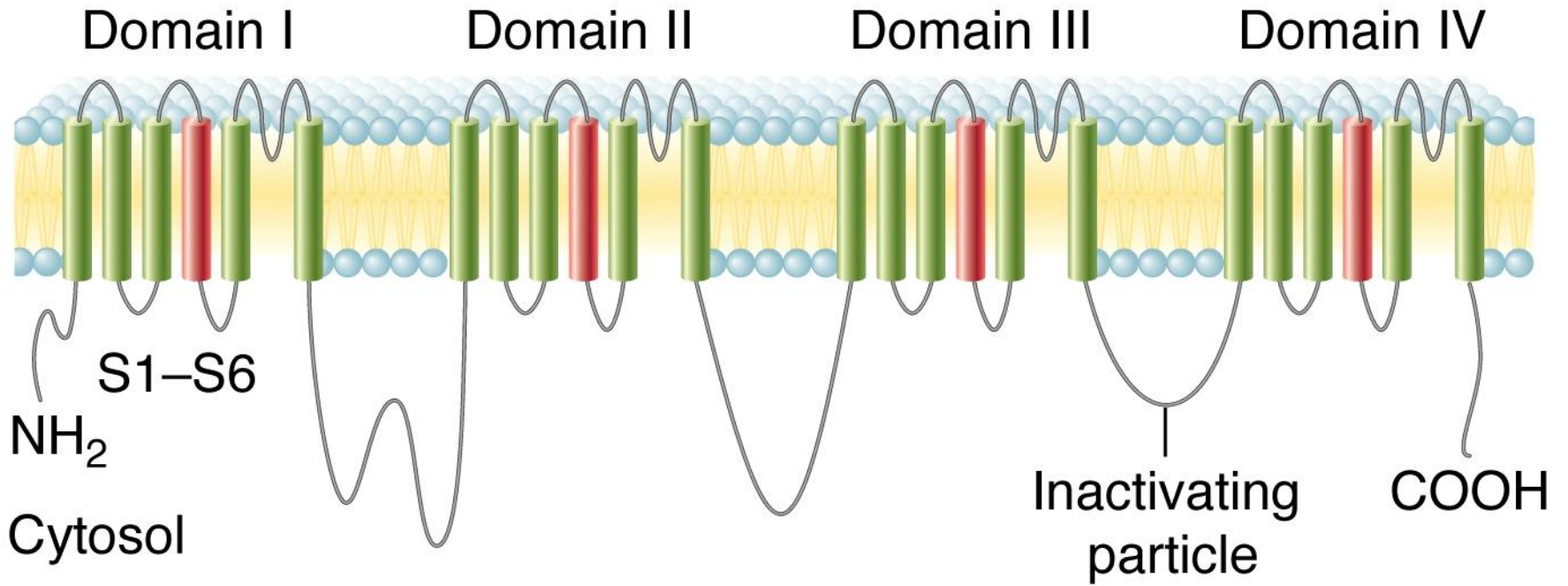


(b)



(a)

Extracellular fluid

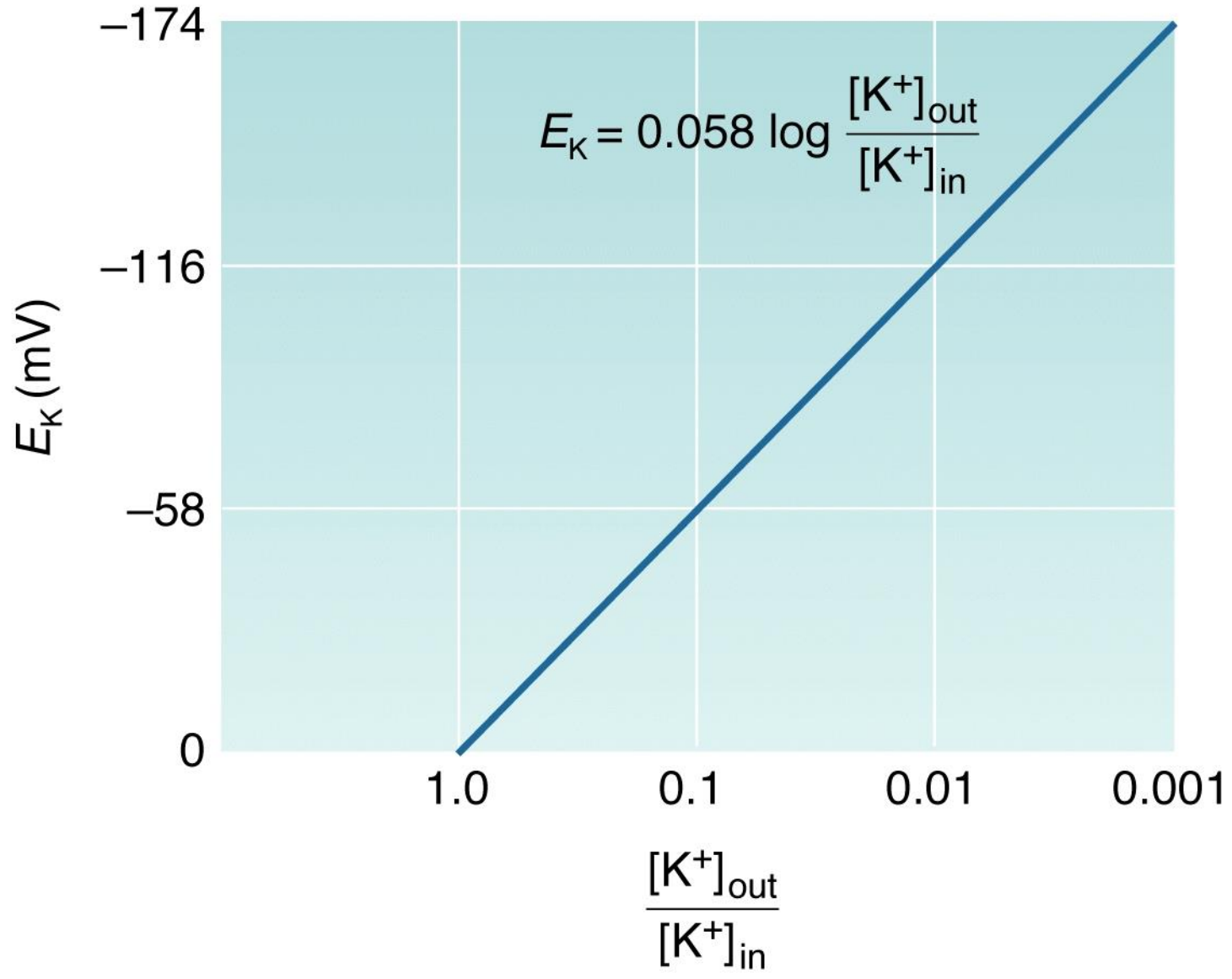


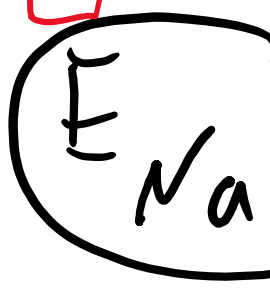
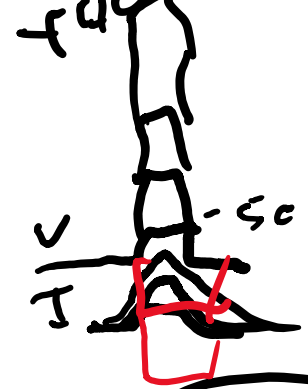
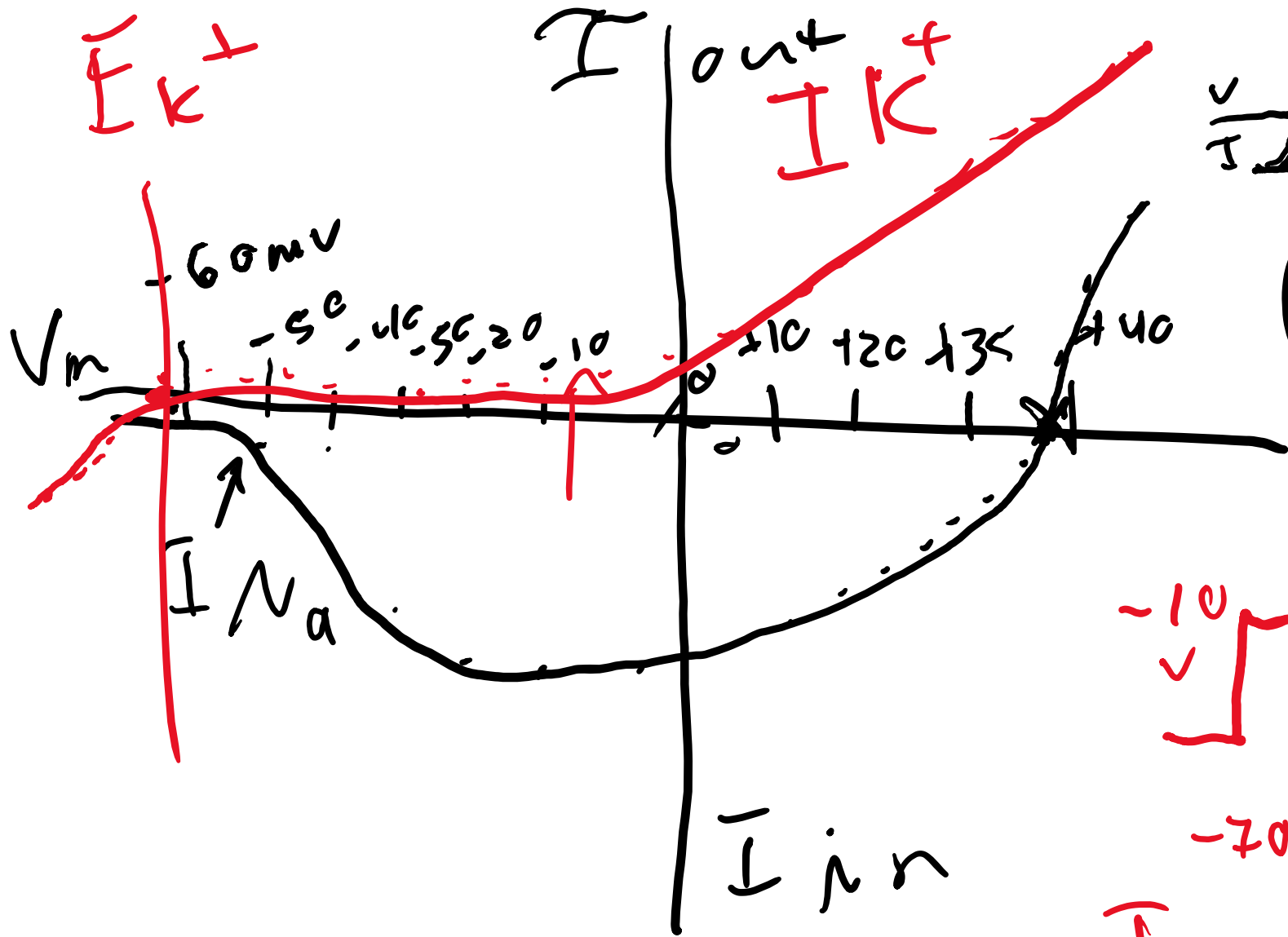
Nernst equation: 1864 – 1941. **Walther Nernst**

$$E = \frac{RT}{zF} \ln \frac{[\text{ion outside cell}]}{[\text{ion inside cell}]} = 2.3026 \frac{RT}{zF} \log_{10} \frac{[\text{ion outside cell}]}{[\text{ion inside cell}]}$$

At room temperature (25 °C), $\frac{RT}{F}$ may be treated like a constant and replaced by 25.693 mV for cells.

(a)





https://www.physiologyweb.com/calculators/ghk_equation_calculator.html